

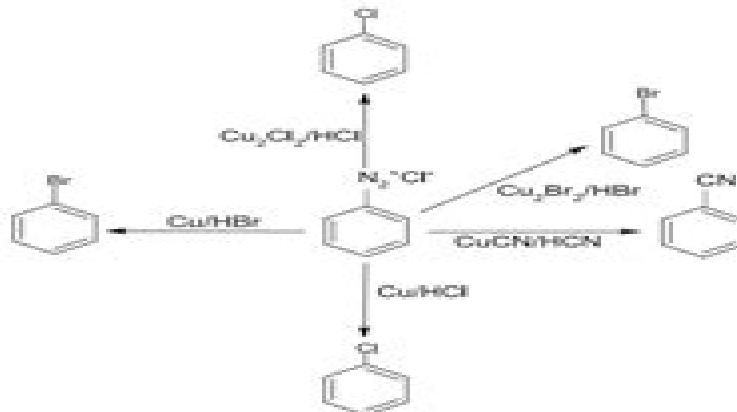
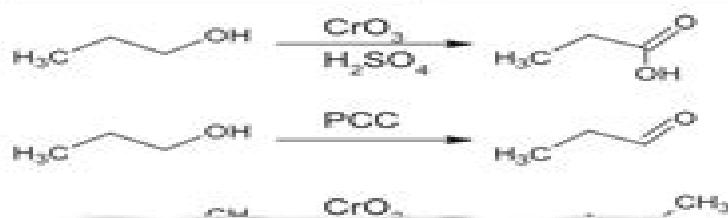
6. CrO_3

Chromium trioxide

- The mixture of chromium trioxide and sulfuric acid dissolved in a mixture of acetone and water is called **Jones reagent**.
- In aqueous acidic solutions, CrO_3 forms chromic acid (H_2CrO_4) which acts as a strong oxidizing agent.
- It is also used to prepare **Pyridinium chlorochromate (PCC)** (**Pyridine + $\text{CrO}_3 + \text{HCl}$**) which is a mild oxidizing agent.

Function -

Mild oxidizing agent	Strong oxidizing agent
1. converts primary alcohols to Aldehydes	1. converts primary alcohols to carboxylic acids
2. secondary alcohols to ketones.	2. secondary alcohols to ketones.
3. NO oxidation of tertiary alcohols	3. NO oxidation of tertiary alcohols



Sandmeyer's Reaction

$\text{Cu}_2\text{Cl}_2/\text{HCl}$

$\text{Cu}_2\text{Br}_2/\text{HBr}$

CuCN/HCN

Gatterman's Reaction

Cu/HCl

Cu/HBr

- It is also used for preparing organocuprates (Gilman reagents)

Organic Chemistry Reagent Study Guide

Susan McMurry

Organic Chemistry Reagent Study Guide:

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