

NYS EAS Study Guide

30 Page Study Guide



The screenshot shows a web browser window with the NYS EAS Study Guide document open. The browser's address bar shows the URL <https://www.nysed.gov/eas-study-guide>. The document is titled "EAS STUDY GUIDE" and is displayed in a clean, professional layout. The left sidebar contains a table of contents with links to various sections. The main content area is titled "EAS STUDY GUIDE" and includes sections for "TEST FORMAT", "TEST DESIGN & FRAMEWORK", and "PASSING SCORE: 528". The "TEST FORMAT" section details the computer-based test structure, including 40 multiple choice questions and 3 written responses, with a total test time of 2.5 hours. The "TEST DESIGN & FRAMEWORK" section includes a table showing the distribution of questions across different competency areas.

EAS STUDY GUIDE

TEST FORMAT

Computer Based Test

- 40 Multiple Choice Questions and 3 Written Responses
- Multiple Choice: 70% of the total score
- Written Responses: 30% of the total score

2.5 hours (15 mins for set up and 2hrs and 15 mins of test time)

Candidates are free to set their own pace during the test administration. The following estimates were used to determine the total test time:

- Written Response: 10 mins each (30 mins total)
- Multiple choice: 105 mins total (~2.6 minutes per question)

PASSING SCORE: 528

TEST DESIGN & FRAMEWORK

Competency	Selected-Response		Constructed-Response	
	Approximate Number of Items	Approximate Percentage of Test Score	Number of Items	Approximate Percentage of Test Score
0001 Diverse Student Populations	11	18%	1	10%
0002 English Language Learners	11	18%	1	10%
0003 Students with Disabilities and Other Special Learning Needs	10	18%	1	10%

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web reacting two solutions eg acid and alkali place the polystyrene cup inside the glass beaker to make it more stable measure an appropriate volume of each liquid eg 25 cm³ place one of the

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web for reactions involving acetic acid or ammonia the measured enthalpy change of neutralization is a few kJ less exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of sodium hydroxide solution with hcl as 57.9 kJ mol⁻¹

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