

12.11.12 - Energy & Matter (Summer 2016)



Introduction

The purpose of this activity is to understand the relationship between energy change associated with the transformation of matter and the conservation of matter and mass in a closed system.

The following procedure will be used in the experiment:

1. A known amount of water is weighed and placed in a beaker.
2. A known amount of metal is weighed and placed in the beaker.
3. The beaker is placed in a calorimeter and the temperature of the water is recorded.
4. The metal is heated in a Bunsen burner and then placed in the beaker.
5. The temperature of the water is recorded after the metal has cooled.
6. The mass of the water and metal is recorded.
7. The mass of the water and metal is recorded after the reaction has taken place.

The results of the experiment will be used to determine the energy change associated with the transformation of matter and the conservation of matter and mass in a closed system. The results will be used to determine the energy change associated with the transformation of matter and mass in a closed system.

Practical objectives

1. To determine the energy change associated with the transformation of matter and mass in a closed system.
2. To determine the conservation of matter and mass in a closed system.
3. To determine the energy change associated with the transformation of matter and mass in a closed system.

<p>Initial mass of water = 100.0 g</p> <p>Initial mass of metal = 10.0 g</p>	<p>Final mass of water = 100.0 g</p> <p>Final mass of metal = 10.0 g</p>
<p>Initial mass of water = 100.0 g</p> <p>Initial mass of metal = 10.0 g</p>	<p>Final mass of water = 100.0 g</p> <p>Final mass of metal = 10.0 g</p>
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