

# Biochemistry Study Guide

1. Organic molecules
  1. **Macromolecules:** Large complex molecules found only in living organisms.
  2. Define the term organic molecule. **Molecules that ave carbon at the center**
  3. Organic molecules are organized around central carbon skeleton.
2. Hydrocarbons
  1. Carbon atoms have four electron in their valence shell, so they can form up to four covalent bonds with other atoms.
  2. Define the term hydro-carbon. **Organic molecules consisting entirely of hydrogen and carbon**
  3. Why are hydrocarbons hydrophobic and insoluble in water? **Hydrocarbons are non polar due to slightly differing electronegativities between carbon and hydrogen**
  4. Describe the difference between aliphatic hydrocarbons and aromatic hydrocarbons. Be able to recognize the structural formula of each. **Aliphatic contain a conjugated bond system, while aromatic do not**
3. Organic molecules are important for living organisms, because of their complex structure.
  1. Organic molecules can form isomers. Define the term isomer. **A molecule that hase the same molecular formula but in different structural formulas**
  2. Organic molecules can form covalent bonds with functional groups. Define the term functional group. **A group of atoms that are attached to the carbon skeleton of an organic molecule**
  3. Be able to name the seven functional groups listed in Figure 2.27 on p. 53, and to recognize their structure.
  4. Be able to recognize each of the seven functional groups within a larger organic molecule.
  5. Macromolecules can be very large.
  6. Name the four groups of macromolecules. **Carbohydrates, protiens, nucleic acids, and lipids**
  7. Carbohydrates, proteins, and nucleic acids can be very large, because they are **polymers** that are built from monomers.
  8. Monomers are linked together during a **dehydration synthesis**. A water molecule is always a product of a dehydration synthesis.

# Organic And Biochemistry Study Guide

**Katherine J. Denniston, Joseph J.  
Topping, Robert L. Caret**



## Organic And Biochemistry Study Guide:

Introduction to Organic and Biochemistry Peter C. Scott, Morris Hein, 1993-02 Designed for professors who prefer to teach general chemistry topics from one text and organic and biochemistry topics from another this text offers step by step and easy to understand coverage of the important functional groups reactions and macromolecules that are essential for health science students A dynamic full color presentation and numerous applications add to the quality of the presentation Content corresponds to Chapter One and Chapters 21-37 of College Chemistry An Introduction to General Organic and Biochemistry Fifth Edition by the same authors Clarity meticulous accuracy and a step by step approach that students can and do understand have become hallmarks of the Hein authorship This new text is no exception Anticipating student problems before they occur the authors move at a manageable pace offering carefully worked out examples with alternate methods of solution practice problems with answers review of concepts review of key terms and a number of other learning aids to ensure student mastery of important material

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**Organic and Biological Chemistry, Study Guide** John R. Holum, 1996-01-18 Explains recent advances in environmental studies and the molecular basis of life Designed for those in the health care field it focuses solely on organic and biochemistry

**Study Guide for Introduction to General, Organic, and Biochemistry** William M. Scovell, 2004

**Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e** Kenneth W. Raymond, 2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond's text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry

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time and guides them through the topics in a systematic way Each chapter of the text is covered by an introduction a list of review topics section by section study suggestions and questions a list of key terms and a practice exam with worked out answers *Study Guide and Solutions Manual for Fundamentals of General, Organic, and Biological Chemistry* Susan

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Topping,Robert L. Caret,2001-06 The third edition of General Organic and Biochemistry is designed to help undergraduate health related majors and students of all other majors understand key concepts and appreciate the significant connections between chemistry health disease and the treatment of disease This text strikes a ba **Fundamentals of General,**

**Organic, and Biological Chemistry, Eighth Edition** John McMurry,Susan McMurry,David Ballantine,Carl Hoeger, Virginia

Peterson,2016-02-24 This manual provides solutions to all problems in the text It explains in detail how the answers to the in text and end of chapter problems are obtained It also contains chapter summaries study hints and self tests for each chapter

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Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section s critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text

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