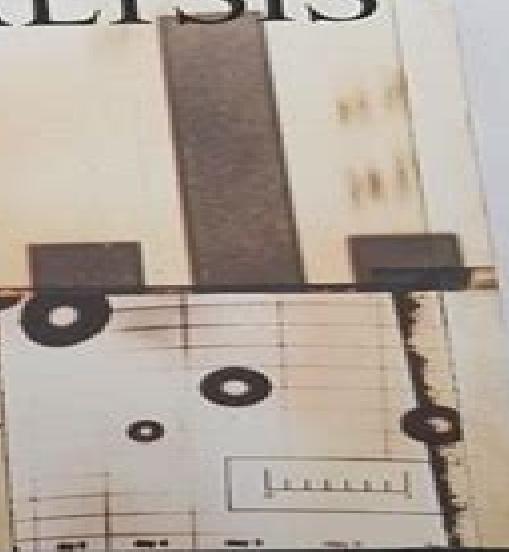
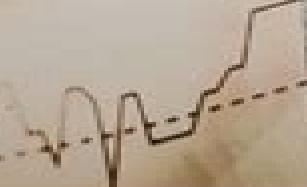
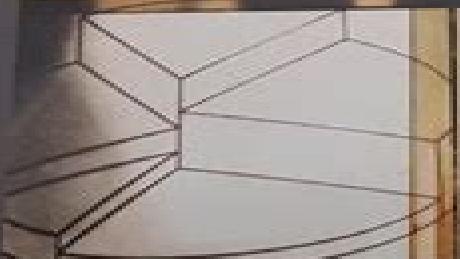


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chemical reaction nh 3 g bf 3 g nh 3 bf 3 s show that the reaction illustrated by this equation is an acid base reaction if we  
use the lewis definitions of an acid and a base see exercise 7

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web kw the product of the concentrations of the hydrogen ions and hydroxide ions acidic solution one which h is greater than the oh basic solution one in which the h is less than the oh alkaline solutions the h of a basic solution is less than  $1 \times 10^{-7}$  m

they are the same as basic solutions

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