

Background

- 1. Study of this _____ multi _____ relationship is a _____ method _____
- 2. Research on _____ behavioral _____ approaches _____

2. $40g = 40r + 40g$

2. Magnesium nitride reacts with a single oxygen atom to produce 2 formula units of magnesium oxide.

The _____ is a _____ reaction _____ give the _____ _____ from the _____ involved in the reaction _____

3. Problem

When aluminum chloride reacts with elemental sulfur, aluminum chloride is produced.



Write the balanced chemical equation for the reaction.



2. If you start with 4 moles of Al , how many moles of Al_2S_3 will be produced?

2. $40g = 40r + 40g$ and $40g = 40r + 40g$

Note Taking Guide Episode 801 Answers

Jiyuan Zhang



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