

**Note Taking Guide: Episode 301 Name\_\_\_\_\_**

**Model:** A familiar idea used to explain unfamiliar facts observed in nature.

**Theory:** An explanation of observable facts and phenomena. To

remain valid, models and theories must:

- explain all known facts
- enable scientists to make correct predictions.

**Democritus:**

- proposed the existence of an atom
- word comes from the greek word atomis which means not to cut or "indivisible"

**Aristotle**

- rejected the idea of the atom
- said matter could be cut continually

**Dalton's theory proposed that atoms:**

- are the building blocks of matter
- are indivisible
- of the same elements are identical
- of different elements are different
- unite in small, whole number ratios to form compounds.

**J.J. Thomson**

- credited with discovery of electron: a blow to Dalton's indivisible atom.
- proposed the plum pudding model of the atom:  
Negatively charged electrons embedded in a ball of positive charge

# Note Taking Guide Episode 1101 Chemistry

**M Carnoy**



**Note Taking Guide Episode 1101 Chemistry:**

*Lovejoy's College Guide ,1993*

## **Note Taking Guide Episode 1101 Chemistry** Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the power of words has are more evident than ever. They have the capability to inspire, provoke, and ignite change. Such may be the essence of the book **Note Taking Guide Episode 1101 Chemistry**, a literary masterpiece that delves deep into the significance of words and their impact on our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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