

A chemical bond - forms when 2 or more atoms rearrange _____ to increase _____

ionic bond - forms when valence _____ are _____ from one atom to another

cation - atom _____ electrons to become _____ charged

anion - atom _____ electrons to become _____ charged

In ionic compounds the ions are arranged in a _____ forces hold the ions together.

Properties of ionic compounds:

- High _____ and _____ points
- _____ - not easily _____
- _____ electricity when _____ or _____ because the ions are free to _____

covalent bond - _____ are _____ forming _____

Covalent compounds have _____ forces holding the _____ together.

Properties of covalent compounds:

- lower _____ and _____ points
- Many covalent compounds are _____ liquids or gases.
- _____ - easier to _____
- come out _____ of electricity

electronegativity - property that tells how strong an atom's _____ is for _____

Since oxygen has a _____ electronegativity than hydrogen, oxygen holds onto shared electrons _____, giving the oxygen a _____ negative charge and the hydrogen a partial _____ charge.

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