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AUTHOR Branden Condyl born Branden Michael Condyl on January 19th 1995 is an American internet celebrity entrepreneur author and social media influencer who has revolutionized the digital marketing industry With over a decade of experience he generated millions in revenue through his innovative social media strategies and motivational content Known for his impactful presence on Instagram Branden has been featured in major publications and has made a significant impact in the entrepreneurship and personal branding spaces He began his entrepreneurial journey in his early twenties achieving notable success by 25 Branden Condyl s influence spans global stages and he has garnered millions of organic social media impressions inspiring individuals worldwide with his unique insights and exceptional communication skills Learn more about Branden Condyl on his website www.brandencondyl.com

Proceedings of the Jepara International Conference on Education and Social Science 2024 (JIC 2024) Agus Subhan Akbar,Mayadina Rohmi Musfiroh,Mochammad Qomaruddin,Mohammad Rifqy Roosdhani,Husni Mubarak,Nina Sofiana,2025-12-16 This is an open access book Jepara International Conference JIC is an international conference held annually in Jepara Indonesia The conference aims to bring together scholars researchers and practitioners from around the world to discuss and share their latest findings on a variety of topics related to sustainable development education and digital society The theme of the 2024 JIC is The Sustainable

Technology of Coastal Development Advance Education and Digital Society This theme is particularly relevant to Jepara a city located on the north coast of Java Island Jepara is facing a number of challenges related to coastal erosion climate change and the need to improve education and digital literacy The JIC will provide a forum for discussing these challenges and identifying innovative solutions The JIC will feature a number of keynote presentations panel discussions and paper presentations There will also be opportunities for networking and collaboration among participants

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the objectives of the boiling point determination lab were to determine the boiling point of a mystery substance provided to each student boiling point is the point at which a liquid changes physical states and becomes a vapor or gas boiling point is important to understand because it can be used to help distinguish between different substances

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at a given pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 °C and 64 °C respectively are far off in a significant way this seems like a valid conclusion. Thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results.

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melting point 1. a. push the open end of a capillary tube into the powdered acetamide. b. move the powder to the closed end of the capillary tube by tapping it on the table. repeat until the powdered acetamide occupies 1–2 mm of the capillary tube end. c.

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discussion and conclusion. The technique of recrystallization was used because it is a simple method to purify a solid compound. The percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific range.

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introduction. In this lab we will be using boiling points, melting points, and infrared spectrometry to identify unknown compounds. Boiling points are affected by three key factors: the first being intermolecular forces, intermolecular forces affect boiling point based on bond strength.

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determination of boiling points. Introduction: the boiling point of a compound is the temperature at which it changes from a liquid to a gas. This is a physical property often used to identify substances or to check the purity of the compound. It

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clean up. Solid and liquid spills immediately. True conclusions after conducting the boiling point determination lab, it can be concluded that the unknown substance 193 is isopropyl alcohol on account of its observed boiling point of 88 °C.

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conclusion after completing this experiment. The melting point of three different unknown compounds was obtained using the melting point apparatus. Compound A had a melting point range of 114–115 °C, thus revealing it to be acetanilide. Compound B had a melting point range of 122–124 °C, thus

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nov 27 2019 essay sample the purpose of this experiment is to determine the boiling points of various organic compounds and to use these to identify unknowns equipment free essays

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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aug 11 2020 boiling points are very sensitive to changes in applied pressure so all boiling points should be reported with the measured pressure a compound's normal boiling point refers to its boiling point at a pressure of 760 text mm ce hg

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the purpose of the experiment is to understand what vapour pressure is and how it changes with temperature to understand the relationship between vapour pressure at a given temperature and boiling temperature at a given pressure to determine the vapour pressure of a pure liquid at various temperatures

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exploring boiling points abstract in this experiment we are given a set of straight chain alkanes and a set of compounds to compare by following the trendline of the straight chain alkanes it could be concluded that as the molar mass of the straight chain alkanes increases their boiling points increase as well

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laboratory partners samantha and mackenzie goals □ correctly determine the boiling point of an unknown liquid □ measure the refractive index of an unknown □ use the gas chromatogram to determine the mass percentage of a mixture and

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50 c so in order to convert an experimental measurement taken in calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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introduction in chemistry one may encounter an unknown substance that needs to be determined by physical properties such

as boiling point melting point density solubility etc in this experiment i am to figure out the boiling point of a mystery substance in order to identify it chemical responsibility this experiment requires the handling of a bunsen burner boiling 6 2b [step by step procedures for boiling point determination](#) - Mar 15 2023

there are a variety of methods by which a sample s boiling point can be determined including distillation reflux and by using a thiele tube the most straightforward method uses a thiele tube and has the advantage of using less than 0 5ml 0 5 ml of material

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