

**Unit 1 Study Guide — ANSWER KEY**

1. Define scientific method: **a procedure for collecting, testing, and communicating information**
2. Define hypothesis: **a suggested answer to the question, an educated guess**
3. Define experiment: **a scientific test so see what will happen**
4. Define dependent variable: **the variable that you do NOT control. (on a graph is usually on the Y axis)**
5. Define independent variable: **the variable that I control. (on a graph is usually on the X axis)**
6. Define control: **the part of the experiment where everything is left the same.**
7. Define observation: **information collected with any of your 5 senses**
8. Define density: **the ratio of mass per unit volume (D = m/v)**
9. Define accuracy: **the measure of how close your data is to the ACTUAL or true value.**
10. Define precision: **a measure of how close your data (measurements) are to each other.**
11. Why do scientists use SI units of measurement? **So that measurements can be understood internationally (around the world).**

12. List (in the correct order) the steps of the scientific method.

- |                        |                                   |
|------------------------|-----------------------------------|
| 1. make an observation | 4. test hypothesis (experiment)   |
| 2. ask a question      | 5. analyze data/form a conclusion |
| 3. form a hypothesis   | 6. develop a theory               |

13. What mnemonic device is used to remember the metric system?

**K H Da B d c m**  
**g, m, L**

14. What is the sentence (mnemsonic) that will help you remember the METRIC prefixes in order?

**King Henry Died By Drinking Chocolate Milk**

15. Fill in the table below describing the metric system:

	<b>Kilo</b>	<b>Hecto</b>	<b>Deca</b>	<b>Base</b>	<b>Deci</b>	<b>Centi</b>	<b>Milli</b>
Abbreviation	<b>K</b>	<b>H</b>	<b>Da</b>	<b>g, m, L</b>	<b>d</b>	<b>c</b>	<b>m</b>
Meaning in powers of 10	<b><math>10^3</math></b>	<b><math>10^2</math></b>	<b><math>10^1</math></b>	<b><math>10^0</math></b>	<b><math>10^{-1}</math></b>	<b><math>10^{-2}</math></b>	<b><math>10^{-3}</math></b>

**Convert the following:**

- |                                 |                                    |
|---------------------------------|------------------------------------|
| 16. 25L to <u>25000</u> mL      | 21. 25 cm to <u>0.25</u> m         |
| 17. 2.3 km to <u>230,000</u> cm | 22. 56784 mm to <u>0.056784</u> km |
| 18. 0.045 g to <u>45</u> mg     | 23. 15 g to <u>0.015</u> kg        |
| 19. 0.35 kg to <u>350</u> g     | 24. 3.452 kL to <u>3452</u> L      |
| 20. 300 mL to <u>0.300</u> L    | 25. 47 m to <u>47000</u> mm        |

**Scientific Notation:**

Convert the following to scientific notation:

- |            |                                      |            |  |
|------------|--------------------------------------|------------|--|
| 26) 0.005  | <u><math>5 \times 10^{-3}</math></u> | 30) 0.025  | <u><math>2.5 \times 10^{-2}</math></u> |
| 27) 5.050  | <u><math>5.05 \times 10^0</math></u> | 31) 0.0025 | <u><math>2.5 \times 10^{-3}</math></u> |
| 28) 0.0008 | <u><math>8 \times 10^{-4}</math></u> | 32) 500    | <u><math>5 \times 10^2</math></u>      |
| 29) 1,000  | <u><math>1 \times 10^3</math></u>    | 33) 5,000  | <u><math>5 \times 10^3</math></u>      |

# Pbs Unit 5 Study Guide Answer Key

**Mike Jess**



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## **Table of Contents Pbs Unit 5 Study Guide Answer Key**

1. Understanding the eBook Pbs Unit 5 Study Guide Answer Key
  - The Rise of Digital Reading Pbs Unit 5 Study Guide Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Pbs Unit 5 Study Guide Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Pbs Unit 5 Study Guide Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Pbs Unit 5 Study Guide Answer Key
  - Personalized Recommendations
  - Pbs Unit 5 Study Guide Answer Key User Reviews and Ratings
  - Pbs Unit 5 Study Guide Answer Key and Bestseller Lists
5. Accessing Pbs Unit 5 Study Guide Answer Key Free and Paid eBooks
  - Pbs Unit 5 Study Guide Answer Key Public Domain eBooks
  - Pbs Unit 5 Study Guide Answer Key eBook Subscription Services
  - Pbs Unit 5 Study Guide Answer Key Budget-Friendly Options
6. Navigating Pbs Unit 5 Study Guide Answer Key eBook Formats

- ePub, PDF, MOBI, and More
- Pbs Unit 5 Study Guide Answer Key Compatibility with Devices
- Pbs Unit 5 Study Guide Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Pbs Unit 5 Study Guide Answer Key
  - Highlighting and Note-Taking Pbs Unit 5 Study Guide Answer Key
  - Interactive Elements Pbs Unit 5 Study Guide Answer Key
- 8. Staying Engaged with Pbs Unit 5 Study Guide Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Pbs Unit 5 Study Guide Answer Key
- 9. Balancing eBooks and Physical Books Pbs Unit 5 Study Guide Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Pbs Unit 5 Study Guide Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Pbs Unit 5 Study Guide Answer Key
  - Setting Reading Goals Pbs Unit 5 Study Guide Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Pbs Unit 5 Study Guide Answer Key
  - Fact-Checking eBook Content of Pbs Unit 5 Study Guide Answer Key
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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