

1. For 2-pentyne, which of the following describes the orbital overlap of the C2—C3 sigma bond?



- A. $sp\text{—}sp$
- B. $sp^2\text{—}sp^2$
- C. $sp^2\text{—}sp^3$
- D. $p\text{—}p$
- E. $sp^2\text{—}sp$

Answer: A

Learning Objective: 9.1 Describe the structure of alkynes and the nature of their chemical function, explaining how alkynes are named

Difficulty: Easy

2. For 2,4-hexadiyne, which of the following describes the orbital overlap of the C3—C4 sigma bond?



- A. $sp\text{—}sp$
- B. $sp^2\text{—}sp^2$
- C. $sp^2\text{—}sp^3$
- D. $p\text{—}p$
- E. $sp^2\text{—}sp$

Answer: A

Learning Objective: 9.1 Describe the structure of alkynes and the nature of their chemical function, explaining how alkynes are named

Difficulty: Easy

3. The sigma bond of an alkyne is formed from the overlap of which two orbitals?

- A. $sp^2\text{—}sp^3$
- B. $p\text{—}p$
- C. $sp^2\text{—}sp^2$
- D. $s\text{—}s$
- E. $sp\text{—}sp$

Answer: E

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