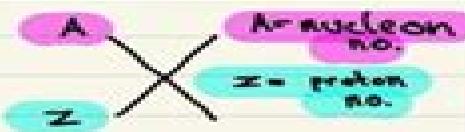
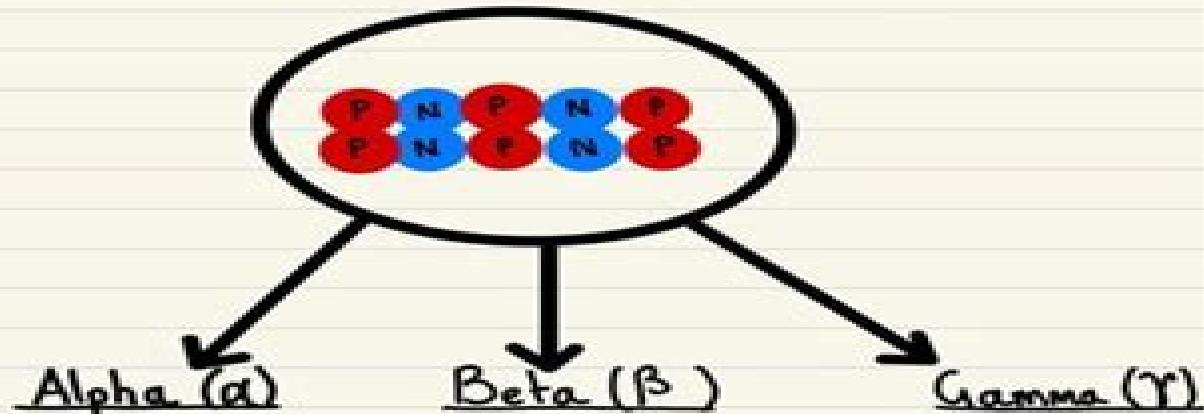


Nuclear Physics



- The number of protons defines an element.
- In an isotope, the number of neutrons is different.
- This means that the nucleon number is different.
- Isotopes with an unbalanced no. of protons and neutrons are more likely to be unstable.
- All radiation is ionising.
- It can remove electrons from atoms (e.g. DNA)
- This causes mutation and can lead to cancer.



- Identical to Helium nucleus. (2P, 2N)
- Can be stopped by thin sheet of paper/box or of air
- Very ionising, does most damage under skin.
- Fast, high energy electron.
- Few mm of aluminium, 1m of air.
- Medium
- High energy EM wave
- Several cm of lead or very thick concrete
- Low ionisation - easily passes through body

Background Radiation

- Naturally occurring e.g. rocks, soil, space (62%)
- Artificial e.g. medical, nuclear power, food + drink + weapons testing.

Nuclear Physics 11 Notes

S Ben Porath

Nuclear Physics 11 Notes:

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