

A. Structural bond: Forces which are more intense than those in the bond
between the electrode and the substrate

Substrate: Forces which include adhesion and cohesion
 from the bond to substrate

Adhesive: an inter atomic attraction to the substrate surface

Substrate: an inter atomic attraction to the adhesive surface

B. Covalent compounds that have are arranged in a crystal lattice
structure of atoms that have regular

Properties of covalent compounds:

- High melting point and boiling points
- hard and strong solid
- soluble in solvents which are not polar
solvents include the hydrocarbons and alcohols

covalent bond: atoms are joined by sharing
electrons

Covalent compounds have atoms joined by
electrons together

Properties of covalent compounds:

- low melting and boiling points
- Many soluble in solvents which are polar
- soft and weak solids
- are not soluble in water

electrochemical: property that allows some strong oxidation and reduction
reactions

Electrochemical is high electrochemical reactions which involve electron
transfer between atoms and ions in the system is called
electrochemistry and the hydrogen peroxide reaction

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