

| Q. No. | Question | |
|--------|---|--|
| 1 | Q. What is the co-ordination no. of Simple Cubic crystal structure. A) 4 B) 6 C) 8 D) 12 | |
| 2 | Q. Which one of the following is covalent crystal A) dry ice B) rock salt C) ice D) quartz | |
| 3 | Q. Points which shows position of atoms in a crystal are called A) lattice point B) lattice line C) lattice circle D) lattice arrangement | |
| 4 | Q. Crystal lattice is also known as. A) lattice triangle B) space lattice C) lattice line D) lattice array | |
| 5 | Q. The number of close neighbor in a body centered cubic lattice of identical sphere is. A) 6 B) 12 C) 4 D) 8 | |
| 6 | Q. Which of the following is a example of ionic crystal solid A) LiF B) Diamond C) Li D) Silicon | |
| 7 | Q. Co-ordination number in Face Centered cubic crystal structure: A) 12 B) 8 C) 6 D) 4 | |
| 8 | Q. On passing electricity through compound it acts as A) Conductor B) Non-conductor C) insulator D) None | |
| 9 | Q. Type of the cubic cell in the given structure A) Hexagonal B) Body-centered cubic C) Face centered cubic D) Simple cubic | |
| 10 | Q. The number of atoms per unit cell of BCC A) 1 B) 2 C) 6 D) 7 | |



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| | |
|----|--|
| 11 | The number of atoms present in molecule indicates its— A) Atomicity B) Acidity C) Neutrality D) Alkalinity |
| 12 | Q. NaCl is an example of——— A) Simple cubic B) Hexagonal close packed C) Body centered cubic D) Face centered cubic |
| 13 | Q. The example of hexagonal closed packed structure is A) Be B) Mg C) Ti D) Al |
| 14 | Q. The HCP structure has..... layers. A) 3 B) 2 C) 1 D) 4 |
| 15 | Q. On which factor conductance of metals responsible? A) ions B) delocalized C) atomic kernel D) number of atoms |
| 16 | Q. Amorphous solids are also referred as.—— A) hard B) supercooled liquids C) rigid D) isotropic |
| 17 | Q. Which of the following is ionic solid? A) Diamond B) sodium chloride C) graphite D) Iodine |
| 18 | Q. Which of the following element become stable after losing an electron A) helium B) Iodine C) sodium D) Oxygen |
| 19 | Q. Which of the following sentence is incorrect for covalent bond ? A) Strength of covalent bond depends upon overlapping of atomic orbitals. B) Covalent bond is not directional C) There is sharing of electrons between atoms bonded by covalent bond D) Covalent bond is formed between atoms having less difference in their electronegativity. |
| 20 | Q. Bonds formed by unequal sharing of electrons are called as A) Non polar Covalent bond B) Polar ionic bond C) Non polar ionic bond D) Polar Covalent bond |

Multiple Choice Questions Of Solid State Chemistry

Arshad Iqbal



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