

Diffusion and Osmosis Worksheet

Diffusion is the movement of particles from areas of higher concentration to areas of lower concentration. It is a natural, random process. This means that it does not require extra energy input.

1a. These are pictures of molecules frozen at two different times. Draw arrows to show the direction each particle might travel due to diffusion in diagrams A and B.



1b. Describe the way the particles of a gas move.

Randomly

1c. Diagram A shows the same number of particles as in Diagram B except most particles start out on one side of the box. Explain why after a while, Diagram A resembles Diagram B.

Molecules move randomly and reach equilibrium

2. Give an everyday example of diffusion in air and in water.

AIR: **e.g. Scents and Smells**

WATER: **e.g. Colored dye in a glass of water**

3. Complete the following sentence: Diffusion is the **-Random-** movement of particles from **-High-** concentration to a **-Low-** concentration until they are **Evenly spread out**.

4. Moths emit chemicals called pheromones to attract a mate.

a) Which process is responsible for the distribution of these chemicals through the air?

Diffusion

b) If a moth detects pheromones, how might it work out which direction they are coming from?

Detect the higher concentration of pheromones

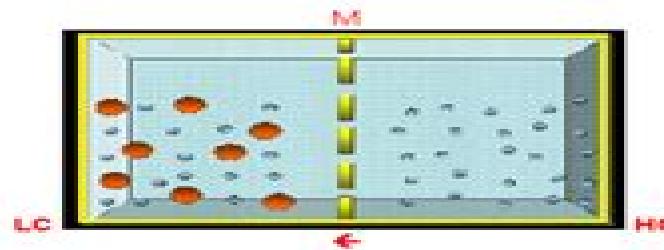
c) What would be the advantage of a moth releasing the pheromones on a night when there was little wind?

The pheromones aren't blown away from a potential mate

5a. Osmosis is the diffusion of water across a semi permeable membrane. The semi permeable membrane acts like a filter that lets only the water through. Water always goes from the area of higher water concentration to the area of lower water concentration. Confused? Then imagine the particles (or solute) are salt and follow the rule - **SALT SUCKS!**

On the diagram on the next page label the:

- Semi permeable membrane (M)
- Area of higher water concentration (HC)
- Area of lower water concentration (LC)
- Direction of osmotic flow (arrow)



Osmosis Study Guide Answer Key

Kim Cooper, Kelly Gosnell

Osmosis Study Guide Answer Key:

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Table of Contents Osmosis Study Guide Answer Key

1. Understanding the eBook Osmosis Study Guide Answer Key
 - The Rise of Digital Reading Osmosis Study Guide Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Osmosis Study Guide Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Osmosis Study Guide Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Osmosis Study Guide Answer Key

- Personalized Recommendations
- Osmosis Study Guide Answer Key User Reviews and Ratings
- Osmosis Study Guide Answer Key and Bestseller Lists

5. Accessing Osmosis Study Guide Answer Key Free and Paid eBooks
 - Osmosis Study Guide Answer Key Public Domain eBooks
 - Osmosis Study Guide Answer Key eBook Subscription Services
 - Osmosis Study Guide Answer Key Budget-Friendly Options
6. Navigating Osmosis Study Guide Answer Key eBook Formats
 - ePUB, PDF, MOBI, and More
 - Osmosis Study Guide Answer Key Compatibility with Devices
 - Osmosis Study Guide Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Osmosis Study Guide Answer Key
 - Highlighting and Note-Taking Osmosis Study Guide Answer Key
 - Interactive Elements Osmosis Study Guide Answer Key
8. Staying Engaged with Osmosis Study Guide Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Osmosis Study Guide Answer Key
9. Balancing eBooks and Physical Books Osmosis Study Guide Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Osmosis Study Guide Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Osmosis Study Guide Answer Key
 - Setting Reading Goals Osmosis Study Guide Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Osmosis Study Guide Answer Key

- Fact-Checking eBook Content of Osmosis Study Guide Answer Key
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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