

# AP Chemistry Pacing Guide

<b>Unit 4: Chemical Reactions</b> <ul style="list-style-type: none"> <li>- Introduction for Reactions</li> <li>- Net Ionic Equations</li> <li>- Representations of Reactions</li> <li>- Physical and Chemical Changes</li> <li>- Stoichiometry</li> <li>- Introduction to Titration</li> <li>- Types of Chemical Reactions</li> <li>- Introduction to Acid-Base Reactions</li> <li>- Oxidation-Reduction (Redox) Reactions</li> </ul>	<b>BIG IDEA 1c:</b> Scale, Proportion, and Quantity <b>BIG IDEA 3c:</b> Transformations	<ol style="list-style-type: none"> <li>I can identify evidence of chemical and physical changes in matter.</li> <li>I can represent changes in matter with a balanced chemical or net ionic equation: a. For physical changes. b. For given information about the identity of the reactants and/or product.</li> <li>For ionic in a given chemical reaction I can represent a given chemical reaction or physical process with a consistent particulate model.</li> <li>I can explain the relationship between macroscopic characteristics and bond interactions for: a. Chemical processes. b. Physical processes.</li> <li>I can explain changes in the amounts of reactants and products based on the balanced reaction equation for a chemical process.</li> <li>I can identify the equivalence point in a titration based on the amounts of the titrant and analyte, assuming the titration reaction goes to completion.</li> <li>I can identify a reaction as acid-base, oxidation-reduction, or precipitation.</li> <li>I can identify species as Brønsted-Lowry acids, bases, and/or conjugate acid-base pairs.</li> <li>I can represent a balanced redox reaction equation using half-reactions.</li> </ol>	10 day(s)
<b>Unit 5: Kinetics</b> <ul style="list-style-type: none"> <li>- Reaction Rates</li> <li>- Introduction to Rate Law</li> <li>- Concentration Changes Over Time</li> <li>- Elementary Reactions</li> <li>- Collision Model</li> <li>- Reaction Energy Profile</li> <li>- Introduction to Reaction Mechanisms</li> <li>- Mechanisms Reaction Mechanism and Rate Law</li> <li>- Steady State</li> </ul>	<b>BIG IDEA 3c:</b> Transformations <b>BIG IDEA 4i:</b> Energy	<ol style="list-style-type: none"> <li>I can explain the relationship between the rate of a chemical reaction and experimental parameters.</li> <li>2. I can represent experimental data with a consistent rate law expression.</li> <li>I can identify the rate law expression of a chemical reaction using data that show how the concentrations of reaction species change over time.</li> <li>I can represent an elementary reaction as a rate law expression using stoichiometry.</li> <li>I can explain the relationship between the rate of an elementary reaction and the frequency, energy, and orientation of molecular collisions.</li> <li>I can represent the activation energy and overall energy change in an elementary reaction using a reaction energy profile.</li> <li>I can identify the components of a reaction mechanism.</li> <li>I can identify the rate law for a reaction from a mechanism in which the first step is rate limiting.</li> <li>I can identify the rate law for a reaction from a mechanism in which the</li> </ol>	10 day(s)

# Metro Nashville Chemistry Pacing Guide

**Yicheng Fang**

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**The Tennessee Conservationist** ,1952    Energy Research Abstracts ,1986    *Plunkett's Chemicals, Coatings & Plastics Industry Almanac* Jack W. Plunkett,2008-07 Market research guide to the chemicals coatings and plastics industry a tool for strategic planning employment searches or financial research Contains trends analysis statistical tables and an industry glossary Includes one page profiles of 400 leading chemicals coatings and plastics industry firms includes addresses phone numbers executive names    *The Magazine of Wall Street and Business Analyst* ,1970    **Water Resources Research Catalog** ,1969 Beginning with vol 9 only new and continuing but modified projects are listed Vols 8 should be kept as a record of continuing but unchanged projects    **Peterson's Guide to Graduate and Professional Programs, an Overview** ,1995    **ASEE 1995-1996 Profiles of Engineering & Engineering Technology Colleges** ,1997    *The University of Tennessee Record* University of Tennessee,1961

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