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remember that the two 94 neutrons \uparrow pp 100 40 each contribute

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web in balancing a nuclear equation it is important to remember that the sum of all the mass numbers and atomic numbers given on the upper left and lower left side of the element symbol respectively must be equal for both sides of the equation

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 $\text{10 o}_2 \text{ co}_2 \text{ h}_2 \text{ o nh}_3 \text{ cuo cu n}_2 \text{ h}_2 \text{ o q10 5 5}$ balance the following equations $\text{fe s cl}_2 \text{ g fecl}_3 \text{ g c}_4 \text{ h}_10 \text{ o}_2 \text{ co}_2 \text{ h}_2 \text{ o as}$
 $\text{naoh na}_3 \text{ aso}_3 \text{ h}$

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web nuclear reactions are going on all around us using correctly balanced equations is important when trying to understand nuclear reactions all equations need to be balanced to conform to two conservation laws the mass number is conserved and the electrical charge of the nucleus is conserved