

Required Information

Use the following information for the Problems below. (Allgo)

[The following information applies to the questions displayed below.]

Forten Company's current year income statement, comparative balance sheets, and additional information follow. For the year, (1) all sales are credit sales, (2) all credits to Accounts Receivable reflect cash receipts from customers, (3) all purchases of inventory are on credit, and (4) all debits to Accounts Payable reflect cash payments for inventory.

FORTEN COMPANY Income Statement For Current Year Ended December 31		
Sales		\$ 422,500
Cost of goods sold		<u>250,000</u>
Gross profit		\$ 172,500
Operating expenses (excluding depreciation)	\$ 148,000	
Depreciation expense	<u>38,750</u>	148,000
Other gains (losses)		
Loss on sale of equipment		<u>(15,125)</u>
Income before taxes		\$ 147,225
Income taxes expense		<u>35,000</u>
Net income		<u>\$ 112,225</u>

FORTEN COMPANY Comparative Balance Sheets December 31		
	Current Year	Prior Year
Assets		
Cash	\$ 43,000	\$ 40,000
Accounts receivable	77,000	58,000
Inventory	287,000	219,000
Prepaid expenses	<u>3,200</u>	<u>2,000</u>
Total current assets	409,000	409,000
Equipment	149,500	118,000
Accumulated depreciation—Equipment	<u>(48,000)</u>	<u>(30,000)</u>
Total assets	<u>\$ 509,500</u>	<u>\$ 497,000</u>
Liabilities and Equity		
Accounts payable	\$ 43,000	\$ 126,000
Long-term notes payable	<u>73,000</u>	<u>98,000</u>
Total liabilities	116,000	141,000
Equity		
Common stock, \$5 per value	174,750	150,000
Paid-in capital in excess of par, common stock	<u>49,500</u>	<u>0</u>
Retained earnings	<u>178,750</u>	<u>106,000</u>
Total liabilities and equity	<u>\$ 509,500</u>	<u>\$ 497,000</u>

Additional Information on Current Year Transactions

- The loss on the cash sale of equipment was \$15,125 (details in d).
- Sold equipment costing \$70,875, with accumulated depreciation of \$38,125, for \$75,625 cash.
- Purchased equipment costing \$104,375 by paying \$46,000 cash and signing a long-term notes payable for the balance.
- Paid \$49,325 cash to reduce the long-term notes payable.
- Issued 3,500 shares of common stock for \$20 cash per share.
- Declared and paid cash dividends of \$5,700.

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RJ Alexander



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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95 82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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