

# *Mark V ProVis®*

## Service Manual

KMP 870

CE 0095

98411-T-205 Rev. D



# Medrad Mark Iv Service Manual

**British Motor Corporation**



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web 1 in the given mixture the sum of all the mole fractions is equal to one  $\sum x_i = 1$  2 the multiplication of the mole fraction by 100 gives the mole percentage 3 mole fraction is a unitless and dimensionless expression advantages of molar fraction mole fraction is not dependent on the temperature

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web nov 6 2023  $x_a = \frac{\text{mol } a}{\text{mol } a + \text{mol } b}$  and  $x_b = \frac{\text{mol } b}{\text{mol } a + \text{mol } b}$  note in the given mixture the sum of all the mole fractions present is equal to one  $x_a + x_b = 1$  when the mole fractions are multiplied by 100 they give the mole percentage mole fraction is a unitless and dimensionless expression image will be uploaded soon

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