

**PN HESI EXIT EXAM 2023 VERSION 1, 2, 3  
TEST BANK QUESTION WITH ANSWERS LATEST UPDATE  
A+ GRADE**

1. During admission to the psychiatric unit, a female client is extremely anxious and states that she is worried about the sun coming up the next day. What intervention is most important for the RN to implement during the admission process?
  - A. **Assist the client in developing alternative coping skills.**
  - B. Remain calm and use a matter-of-fact approach.
  - C. Ask the client why she is so anxious.
  - D. Administer a PRN sedative to help relieve her anxiety.
  
2. A female client is brought to the emergency department after police officers found her disoriented, disorganized, and confused. The RN also determines that the client is homeless and is exhibiting suspiciousness. The client's plan of care should include what priority problem?
  - A. **Acute confusion.**
  - B. Ineffective community coping.
  - C. Disturbed sensory perception.
  - D. Self-care deficit.
  
3. The occupational health nurse is working with a female employee who was just notified that her child was involved in a MVA and taken to the hospital. The employee states, "I can't believe this. What should I do?" Which response is best for the RN to provide in this crisis?
  - A. Tell me what you think should happen.
  - B. How serious was the collision?
  - C. What do you think you should do?
  - D. **Call for transportation to the hospital.**
  
4. A client tells the RN that he has an IQ of 400+ and is a genius and an inventor. He also reports that he is married to a female movie star and thinks that his brother wants a sexual relationship with her. What is the priority nursing problem for admission to the psychiatric unit?
  - A. **Ineffective sexual patterns.**
  - B. Impaired environmental interpretation.
  - C. Disturbed sensory perception.
  - D. Compromised family coping.

# Pdf Hesi Exit Exam Test Bank

**Garrett Hamm**



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web elements in the second row of the periodic table place their electrons in the 2n shell as well as the 1n shell for instance lithium li has three electrons two fill the 1 s orbital and the third is placed in the 2 s orbital giving an electron configuration of 1 s 2 2 s 1

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