

# UNE Medical Biochemistry Exam 1 Questions With Complete Solutions.

what percentage of water is in the body?

60%

**Relationship between protons and acids & bases**

Acids donate proton. Bases accept proton

**Strong Acids Vs Weak Acids**

Strong acids fully dissociate (100%) and weak acids partially dissociate

**properties of water**

dipole dipole

polar

H bonding

dissolves electrolytes ( Cl- /Na+)

Small Kd- doesn't dissociate a lot

**kw ion product**

10^-14

**pH**

-log[H+]

**biological PH**

7.4

**types of buffers**

phosphates, bicarbonate, proteins

**Buffers**

weak acid and conjugate base

**Ka**

how well acid can dissociate

**pka**

-log(Ka)

50% dissociation

**Buffers work best when?**

pH=pka

within +/- 1 pH unit of pka

**Henderson-Hasselbalch equation**

pH= pka + log A-/HA

**pH< pKa**

protonated

**pH> pKa**

ionized

**amino acids are linked together by**

peptide bonds between carboxyl and amino group

**what's special about histidine pka?**

close to biological PH of 7.4 so it can act as an donor or acceptor

nonpolar amino acids

# Medical Biochemistry Exam Questions Une

**JG Myers**

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