

16

SOLUTIONS

SECTION 16.1 PROPERTIES OF SOLUTIONS (pages 471–477)

This section identifies the factors that affect the solubility of a substance and determine the rate at which a solute dissolves.

► Solution Formation (pages 471–472)

Look at Figure 16.1 on page 471 to help you answer Questions 1 and 2.

- Underline the condition that causes sugar to dissolve *faster* in water.
 - as a whole cube or in granulated form?
 - when allowed to stand or when stirred?
 - at a higher temperature or a lower temperature?
- Name three factors that influence the rate at which a solute dissolves in a solvent.
 - agitation (stirring or shaking)
 - temperature
 - particle size
- Is the following sentence true or false? Finely ground particles dissolve more rapidly than larger particles because finer particles expose a greater surface area to the colliding solvent molecules. true

► Solubility (pages 472–473)

- Complete the following table showing the steps in a procedure to determine the total amount of sodium chloride that will dissolve in 100 g of water at 25°C.

Procedure	Amount Dissolved	Amount Not Dissolved
Add 36.0 g of sodium chloride to the water	36.0 g	0.0 g
Add an additional 1.0 g of sodium chloride	0.2 g	0.8 g
Determine the total amount that dissolves	36.2 g	

- The amount of a substance that dissolves in a given quantity of solvent at a constant temperature is called the substance's solubility at that temperature.
- If a solution contains the *maximum* amount of solute for a given quantity of solvent at a constant temperature, it is called a(n) saturated solution.

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Karen Timberlake



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Fundamentals of General, Organic, and Biological Chemistry Susan McMurry, 2006-07 The Study Guide and Full Solutions Manual explain in detail how the answers in text and end of chapter problems are obtained They also contain chapter summaries study hints and self tests for each chapter

Environmental Chemistry Stanley E. Manahan, 2022-06-19 With clear explanations real world examples and updated ancillary material the 11th edition of Environmental Chemistry emphasizes the concepts essential to the practice of environmental science technology and chemistry The format and organization popular in preceding editions is used including an approach based upon the five environmental spheres and the relationship of environmental chemistry to the key concepts of sustainability industrial ecology and green chemistry The new edition provides a comprehensive view of key environmental issues and significantly looks at diseases and pandemics as an environmental problem influenced by other environmental concerns like climate change Features The most trusted and best selling text for environmental chemistry has been fully updated and expanded once again The author has preserved the basic format with appropriate updates including a comprehensive overview of key environmental issues and concerns New to this important text is material on the threat of pathogens and disease deadly past pandemics that killed millions recently emerged diseases and the prospects for more environment threats related to disease This outstanding legacy appeals to a wide audience and can also be an ideal interdisciplinary book for graduate students with degrees in a variety of disciplines other than chemistry New Long awaited companion website featuring additional ancillary material

Working with Text and Around Text in Foreign Language Environments Halina Chodkiewicz, Piotr Steinbrich, Małgorzata Krzemińska-Adamek, 2016-06-28 This book investigates the three pivotal points of text for foreign language acquisition reception construction and deconstruction In Part One the focus is on various aspects of text reception such as developing literacy text interest and perceptions of the academic register or the assessment of spoken language in educational contexts Part Two deals with various aspects of composing text such as author identity lexical constructs or collaborative web based writing Lastly Part Three presents the various segmental items that constitute text like lexical clustering L1 L2 relationship classroom talk as text etc The division corresponds with what can be viewed as a logical sequence of text related processes reflected in formal learning and teaching environments

Guide to the Selection of Books for Your Secondary School Library Dorothy A. McGinniss, 1972

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Growing Up with Science Cavendish Square Publishing LLC, 2006 Volume fifteen of a seventeen volume alphabetically arranged encyclopedia contains approximately five hundred articles introducing key aspects of science and technology

Essential Mathematics and Statistics for Forensic Science Craig Adam, 2011-09-20 This text is an accessible student friendly introduction to the wide range of mathematical and statistical tools needed by the forensic scientist in the analysis interpretation and presentation of experimental measurements From a basis of high school mathematics the book develops

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