

## 2.4 Chemical Reactions and Enzymes

### Lesson Objectives

- ☒ Explain how chemical reactions affect chemical bonds.
- ☒ Describe how energy changes affect how easily a chemical reaction will occur.
- ☒ Explain why enzymes are important to living things.

### Lesson Summary

**Chemical Reactions** Everything that happens in an organism is based on chemical reactions. A **chemical reaction** is a process that changes one set of chemicals into another set of chemicals.

- ▶ The elements or compounds that enter into the reaction are the **reactants**.
- ▶ The elements or compounds produced by the reaction are the **products**.
- ▶ Chemical reactions involve changes in the chemical bonds that join atoms in compounds.

**Energy in Reactions** Some chemical reactions release energy; others absorb energy.

- ▶ Chemical reactions that release energy often occur on their own.
- ▶ Chemical reactions that absorb energy require a source of energy. The energy needed to get a reaction started is called the **activation energy**.

**Enzymes** An **enzyme** is a protein that acts as biological catalyst. A **catalyst** is a substance that speeds up the rate of a chemical reaction. Catalysts work by lowering a reaction's activation energy.

- ▶ In an enzyme-catalyzed reaction, the reactants are known as **substrates**. Substrates bind to a part of an enzyme called the active site and remain bound to the enzyme until the reaction is complete, when the products are released.
- ▶ Temperature, pH, and regulatory molecules can affect the activity of enzymes.

## Chemical Reactions

1. What is a chemical reaction?

*It is a process that changes one set of chemicals into another set of chemicals.*

2. Complete the table about chemicals in a chemical reaction.

Chemicals in a Chemical Reaction	
Chemicals	Definition
Reactants	<i>the elements or compounds that enter into a chemical reaction</i>
Products	<i>the elements or compounds that are produced by a chemical reaction</i>

# Pearson Chemical Reactions Study Guide

**Mark Steyvers**



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## **Table of Contents Pearson Chemical Reactions Study Guide**

1. Understanding the eBook Pearson Chemical Reactions Study Guide
  - The Rise of Digital Reading Pearson Chemical Reactions Study Guide
  - Advantages of eBooks Over Traditional Books
2. Identifying Pearson Chemical Reactions Study Guide
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Pearson Chemical Reactions Study Guide
  - User-Friendly Interface
4. Exploring eBook Recommendations from Pearson Chemical Reactions Study Guide
  - Personalized Recommendations
  - Pearson Chemical Reactions Study Guide User Reviews and Ratings
  - Pearson Chemical Reactions Study Guide and Bestseller Lists

5. Accessing Pearson Chemical Reactions Study Guide Free and Paid eBooks
  - Pearson Chemical Reactions Study Guide Public Domain eBooks
  - Pearson Chemical Reactions Study Guide eBook Subscription Services
  - Pearson Chemical Reactions Study Guide Budget-Friendly Options
6. Navigating Pearson Chemical Reactions Study Guide eBook Formats
  - ePub, PDF, MOBI, and More
  - Pearson Chemical Reactions Study Guide Compatibility with Devices
  - Pearson Chemical Reactions Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Pearson Chemical Reactions Study Guide
  - Highlighting and Note-Taking Pearson Chemical Reactions Study Guide
  - Interactive Elements Pearson Chemical Reactions Study Guide
8. Staying Engaged with Pearson Chemical Reactions Study Guide
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Pearson Chemical Reactions Study Guide
9. Balancing eBooks and Physical Books Pearson Chemical Reactions Study Guide
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Pearson Chemical Reactions Study Guide
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Pearson Chemical Reactions Study Guide
  - Setting Reading Goals Pearson Chemical Reactions Study Guide
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pearson Chemical Reactions Study Guide
  - Fact-Checking eBook Content of Pearson Chemical Reactions Study Guide
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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