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THE BEHAVIOR OF GASES

Chapter Test II

A. Matching

Match a word or phrase in Column I with the correct description in Column II. Write the letter of the correct word or phrase in the space.

Column I

1. The constant relating unit temperature change to unit volume change for a dilute gas at constant pressure (the slope of the straight line of the expansion curve).
2. The volume of a fixed amount of a gas at constant pressure in the standard state (0°C and 1 atm pressure).
3. The ratio of density of a gas to density of water at the same temperature and pressure.
4. Temperature at which a gas exerts a pressure of 1 atm at a volume of 22.4 L.
5. A measure of how much the volume of a gas expands with temperature.
6. The a parameter of gas expansion, temperature, for a fixed amount of gas (mathematically, $\alpha = \frac{1}{V} \left(\frac{\partial V}{\partial T} \right)_P$).
7. An expansion of a gas in which the pressure is constant.
8. $\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$
9. A measure of how much a gas expands with temperature.
10. A gas at 0°C and 1 atm.

Column II

- a. standard gas
- b. ideal gas constant R
- c. density
- d. compressibility
- e. Boyle's law
- f. Boyle's constant
- g. Charles law (mathematical statement)
- h. volume
- i. expansion
- j. coefficient of expansion

B. Multiple Choice

Choose the correct answer and write its letter in the space.

1. Expanding one volume of a substance against one atm of constant pressure requires constant pressure. The volume of the substance is:
a. the same as the initial volume.
b. the same as the initial volume.
c. the same as the initial volume.
d. the same as the initial volume.

Pearson Chemistry Behavior Of Gases

Jiyuan Zhang



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Prentice Hall Chemistry Antony C. Wilbraham, 2008 Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction *Comprehensive Inorganic Chemistry II* ,2013-07-23 Comprehensive Inorganic Chemistry II Nine Volume Set reviews and examines topics of relevance to today's inorganic chemists Covering more interdisciplinary and high impact areas Comprehensive Inorganic Chemistry II includes biological inorganic chemistry solid state chemistry materials chemistry and nanoscience The work is designed to follow on with a different viewpoint and format from our 1973 work Comprehensive Inorganic Chemistry edited by Bailar Emel us Nyholm and Trotman Dickenson which has received over 2 000 citations The new work will also complement other recent Elsevier works in this area Comprehensive Coordination Chemistry and Comprehensive Organometallic Chemistry to form a trio of works covering the whole of modern inorganic chemistry Chapters are designed to provide a valuable long standing scientific resource for both advanced students new to an area and researchers who need further background or answers to a particular problem on the elements their compounds or applications Chapters are written by teams of leading experts under the guidance of the Volume Editors and the Editors in Chief The articles are written at a level that allows undergraduate students to understand the material while providing active researchers with a ready reference resource for information in the field The chapters will not provide basic data on the elements which is available from many sources and the original work but instead concentrate on applications of the elements and their compounds Provides a comprehensive review which serves to put many advances in perspective and allows the reader to make connections to related fields such as biological inorganic chemistry materials chemistry solid state chemistry and nanoscience Inorganic chemistry is rapidly developing which brings about the need for a reference resource such as this that summarise recent developments and simultaneously provide background information Forms the new definitive source for researchers interested in elements and their applications completely replacing the highly cited first edition which published in 1973 *American Gas-light Journal and Chemical Repertory* ,1883 *Long-term Environmental Effects of Offshore Oil and Gas Development* D.F. Boesch,N.N. Rabalais,1987-03-24 Long term Environmental Effects of Offshore Oil and Gas Development contains 14 chapters by different authors which focus on the US *Theoretical and physical chemistry* Samuel Lawrence Bigelow,1912 *Industrial & Engineering Chemistry* ,1924 *Scientific and Technical Aerospace Reports* ,1979 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA Scientific and Technical Information Database **The American Gas Light**

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