

Chemistry Student Edition - Basic Answer Key

Chapter 14: The Properties of Gases

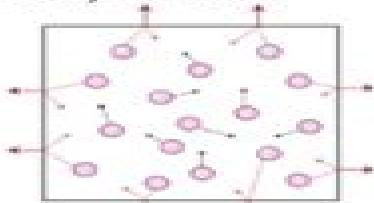
Gas Properties

Questions

1. Why do you suppose the pressure of atmospheric gases increases with proximity to earth? Why aren't the gases evenly distributed throughout the entire reaches of the atmosphere? Study the PhET simulation for further illustration here: <http://phet.colorado.edu/en/simulation/states-of-matter>.
2. Draw a diagram showing how particles exert pressure on the wall of a container.
3. If you were to double the pressure of the bicycle tire from 46 psi to 92 psi, what would be the new pressure in atm?
4. The average atmospheric pressure in Denver is 0.830 atm. Express this pressure in mm Hg.
5. Convert a pressure of 570. torr to atmospheres.
6. Convert 8.5 psi to torr.

Answers

1. The main reason for this is that gravity has an effect on gas particles, just as it does on any other substance that has mass. In the case of gases, they are moving at very high speeds and their collisions are elastic. However, when look at the scale of atmospheric gases, we begin to notice a stratification in gas density. Above 5 miles of altitude, we see a dramatic drop in the amount of gas particles present. At 19 miles, the percent of atmospheric gases approaches zero. Where they are present, however, as in the troposphere, the gases are evenly distributed.



2. The pressure would be double, so $3.1 \text{ atm} \times 2 = 6.2 \text{ atm}$

$$0.830 \text{ atm} \times \left(\frac{760 \text{ mmHg}}{1 \text{ atm}} \right) = 630 \text{ mmHg}$$

4. $570 \text{ torr} \times \left(\frac{1 \text{ atm}}{760 \text{ torr}} \right) = 0.75 \text{ atm}$

$$8.5 \text{ psi} \times \left(\frac{760 \text{ torr}}{14.7 \text{ psi}} \right) = 439 \text{ torr}$$

- 6.

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Chemistry John S. Phillips, Cheryl Wistrom, 2000 **Analysis of Kinetic Data** E. S. Swinbourne, 2013-04-09 Data analysis is important from two points of view first it enables a large mass of information to be reduced to a reasonable compass and second it assists in the interpretation of experimental results against some framework of theory The purpose of this text is to provide a practical introduction to numerical methods of data analysis which have application in the field of experimental chemical kinetics Recognizing that kinetic data have many features in common with data derived from other sources I have considered it appropriate to discuss a selection of general methods of data analysis in the early chapters of the text It is the author's experience that an outline of these methods is not always easy to locate in summary form and that their usefulness is often not sufficiently appreciated Inclusion of these methods in the early chapters has been aimed at simplifying discussion in the later chapters which are more particularly concerned with kinetic systems By the provision of a number of worked examples and problems it is hoped that the reader will develop a feeling for the range of methods available and for their relative merits Throughout the text the mathematical treatment has been kept relatively simple lengthy proofs being avoided I have preferred to indicate the sense and usefulness of the various methods rather than to justify them on strict mathematical grounds **Mathematical Modelling, Optimization, Analytic and Numerical Solutions** Pammy

Manchanda, René Pierre Lozi, Abul Hasan Siddiqi, 2020-02-04 This book discusses a variety of topics related to industrial and applied mathematics focusing on wavelet theory sampling theorems inverse problems and their applications partial differential equations as a model of real world problems computational linguistics mathematical models and methods for meteorology earth systems environmental and medical science and the oil industry It features papers presented at the International Conference in Conjunction with 14th Biennial Conference of ISIAM held at Guru Nanak Dev University Amritsar India on 24 February 2018 The conference has emerged as an influential forum bringing together prominent academic scientists experts from industry and researchers The topics discussed include Schrodinger operators quantum kinetic equations and their application extensions of fractional integral transforms electrical impedance tomography diffuse optical tomography Galerkin method by using wavelets a Cauchy problem associated with Korteweg de Vries equation and entropy solution for scalar conservation laws This book motivates and inspires young researchers in the fields of industrial and applied mathematics *The Physical Basis of Organic Chemistry* Howard Maskill, 1985 This rigorous but not overly mathematical account of the physical principles of modern organic chemistry provides an in depth treatment of the subject not found in general physical or organic chemistry texts The author integrates worked numerical examples throughout as well as including them at the end of each chapter It is appropriate for courses in physical organic chemistry and physical biochemistry at the upper division and graduate level *Analytical Chemistry: Qualitative analysis* Frederick Pearson Treadwell, William Thomas Hall, 1908 *Customized Modern Inorganic Chemistry* William L. Jolly, 1998 **Modern Inorganic Chemistry** William L. Jolly, 1984 **Solutions Manual** Joseph Topich, 2000-10 *Chemistry* Dorin, Prentice Hall Dictionary Editors, 1992 **Remediation of Sites with Dispersed Radioactive Contamination** International Atomic Energy Agency, 2004 Provides an overview of remediation technologies that are particularly suited to the remediation of dispersed contamination The technologies are outlined in brief and their advantages and limitations are discussed The need for a holistic design of the remedial action is stressed *Technical Reports Series*, 2004 **Journal of Reading**, 1986 *Magazine of Concrete Research*, 1977 **Knowledge & Illustrated Scientific News**, 1883 *British Abstracts*, 1935

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