

## Chemistry Student Edition - Basic Answer Key

### Chapter 14: The Properties of Gases

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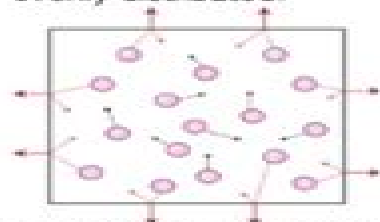
#### Gas Properties

##### Questions

1. Why do you suppose the pressure of atmospheric gases increases with proximity to earth? Why aren't the gases evenly distributed throughout the entire reaches of the atmosphere? Study the PhET simulation for further illustration here: <http://phet.colorado.edu/en/simulation/states-of-matter>.
2. Draw a diagram showing how particles exert pressure on the wall of a container.
3. If you were to double the pressure of the bicycle tire from 46 psi to 92 psi, what would be the new pressure in atm?
4. The average atmospheric pressure in Denver is 0.830 atm. Express this pressure in mm Hg.
5. Convert a pressure of 570. torr to atmospheres.
6. Convert 8.5 psi to torr.

##### Answers

1. The main reason for this is that gravity has an effect on gas particles, just as it does on any other substance that has mass. In the case of gases, they are moving at very high speeds and their collisions are elastic. However, when look at the scale of atmospheric gases, we begin to notice a stratification in gas density. Above 5 miles of altitude, we see a dramatic drop in the amount of gas particles present. At 19 miles, the percent of atmospheric gases approaches zero. Where they are present, however, as in the troposphere, the gases are evenly distributed.



- 2.
3. The pressure would be double, so  $3.1 \text{ atm} \times 2 = 6.2 \text{ atm}$
4. 
$$0.830 \text{ atm} \times \left( \frac{760 \text{ mmHg}}{1 \text{ atm}} \right) = 630 \text{ mmHg}$$
5. 
$$570 \text{ torr} \times \left( \frac{1 \text{ atm}}{760 \text{ torr}} \right) = 0.75 \text{ atm}$$
6. 
$$8.5 \text{ psi} \times \left( \frac{760 \text{ torr}}{14.7 \text{ psi}} \right) = 439 \text{ torr}$$

# Pearson Chemistry Answer Key Ch 14

**RJ Shavelson**



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**Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science**, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Chemistry: The Central Science** Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Dalius Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation **Prentice Hall Chemistry** Antony C. Wilbraham, 2008 Chemistry Pearson Education, Lemay, 2005-05 This lecture notebook contains the art from the text with note taking sections to obviate the need for students to spend time redrawing figures in lecture and instead lets them concentrate on taking notes *Fundamentals of Photochemistry* K. K. Rohatgi-Mukherjee, 1978 *Chemistry* John S. Phillips, Cheryl Wistrom, 2000 **Analysis of Kinetic Data** E. S. Swinbourne, 2013-04-09 Data analysis is important from two points of view first it enables a large mass of information to be reduced to a reasonable compass and second it assists in the interpretation of experimental results against some framework of theory The purpose of this text is to provide a practical introduction to numerical methods of data analysis which have application in the field of experimental chemical kinetics Recognizing that kinetic data have many features in common with data derived from other sources I have considered it appropriate to discuss a selection of general methods of data analysis in the early chapters of the text It is the author's experience that an outline of these methods is not always easy to locate in summary form and that their usefulness is often not sufficiently appreciated Inclusion of these methods in the early chapters has been aimed at simplifying discussion in the later chapters which are more particularly concerned with kinetic systems By the provision of a number of worked examples and problems it is hoped that the reader will develop a feeling for the range of methods available and for their relative merits Throughout the text the mathematical treatment has been kept relatively simple lengthy proofs being avoided I have preferred to indicate the sense and usefulness of the various methods rather than to justify them on strict mathematical grounds *Mathematical Modelling, Optimization, Analytic and Numerical Solutions* Pammy

Manchanda, René Pierre Lozi, Abul Hasan Siddiqi, 2020-02-04 This book discusses a variety of topics related to industrial and applied mathematics focusing on wavelet theory sampling theorems inverse problems and their applications partial differential equations as a model of real world problems computational linguistics mathematical models and methods for meteorology earth systems environmental and medical science and the oil industry It features papers presented at the International Conference in Conjunction with 14th Biennial Conference of ISIAM held at Guru Nanak Dev University Amritsar India on 24 February 2018 The conference has emerged as an influential forum bringing together prominent academic scientists experts from industry and researchers The topics discussed include Schrodinger operators quantum kinetic equations and their application extensions of fractional integral transforms electrical impedance tomography diffuse optical tomography Galerkin method by using wavelets a Cauchy problem associated with Korteweg de Vries equation and entropy solution for scalar conservation laws This book motivates and inspires young researchers in the fields of industrial and applied mathematics

*The Physical Basis of Organic Chemistry* Howard Maskill, 1985 This rigorous but not overly mathematical account of the physical principles of modern organic chemistry provides an in depth treatment of the subject not found in general physical or organic chemistry texts The author integrates worked numerical examples throughout as well as including them at the end of each chapter It is appropriate for courses in physical organic chemistry and physical biochemistry at the upper division and graduate level

Analytical Chemistry: Qualitative analysis Frederick Pearson Treadwell, William Thomas Hall, 1908

Customized Modern Inorganic Chemistry William L. Jolly, 1998

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**Journal of Reading**, 1986

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