

Section Review 8.1

Part A Completion

1. compound	6. low
2. molecular	7. high
3. nonmetals	8. atoms
4. diatomic	9. structure
5. molecular formula	

Part B True-False

10. ST	12. AT	14. ST
11. NT	13. NT	

Part C Matching

15. e	17. d	19. c
16. a	18. b	

Part D Questions and Problems

20. molecular compound
21. a. 4 carbon atoms, 10 hydrogen atoms
b. 6 carbon atoms, 5 hydrogen atoms, 1 fluorine atom
22. a. molecule d. atom
b. atom e. molecule
c. molecule

Section Review 8.2

Part A Completion

- stable electron
- covalent
- shared
- single
- unshared pairs
- double/triple
- coordinate covalent bond
- Energy
- bond dissociation energy
- resonance structure

Part B True-False

11. NT	13. AT	15. ST
12. NT	14. AT	16. NT

Part C Matching

17. e	19. b	21. d
18. a	20. c	

Part D Questions and Problems

22. a. $\text{:Br}-\text{Br}:$
b. $\text{H}-\text{C}=\text{C}-\text{H}$
c. $\begin{array}{c} \text{H} \\ | \\ \text{H}-\text{N}-\text{H} \\ | \\ \text{H} \end{array}$

Section 8.3

Part A Completion

- molecular orbitals
- bonding orbital
- lower
- sigma or σ
- pi or π
- three-dimensional
- VSEPR theory
- orbital hybridization

Part B True-False

9. AT	11. NT	13. ST
10. ST	12. AT	14. AT

Part C Matching

15. d	17. a	19. c
16. e	18. b	

Part D Questions and Problems

20. Reading from left to right:
 sp^3 sp^2 sp^2 sp sp sp^2

Section 8.4

Part A Completion

- equally
- nons polar
- unequally
- polar
- electronegativities
- dipole interactions
- hydrogen bond
- electronegative
- oxygen, nitrogen, or fluorine

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