

PRINCIPLES OF MODERN CHEMISTRY

SEVENTH EDITION

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Carl Hoeger, Laurence Lavelle, Yinfia Ma

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Freeman,2003 **Elektron dalam Atom** Eko Sulistio Hanam,Vina N. Van Harling,Elvinda Bendra Agustina,Andi Budirohmi,Ely Rismawati,Semuel Simra Pada,Dine Agustine,Juniar Afrida,Asep Irvan Irvani,Widiastini Arifuddin,Dhian Eka Wijaya,Syarifuddin,Niken Sylvia Puspitasari,2024-09-14 Elektron dalam Atom adalah buku yang menjelaskan secara mendalam tentang peran dan sifat elektron dalam struktur atom serta bagaimana keberadaannya memengaruhi sifat sifat materi Buku ini mengupas konsep konsep dasar fisika atom termasuk teori kuantum konfigurasi elektron dan ikatan kimia yang menjadi fondasi pemahaman tentang bagaimana atom berinteraksi dan membentuk berbagai bentuk materi Dituliskan dengan pendekatan yang sistematis dan mudah dipahami buku ini menyajikan penjelasan yang komprehensif tentang model model atom dari berbagai era mulai dari model atom klasik hingga teori kuantum modern Pembaca akan diajak untuk memahami konsep konsep kunci seperti energi ionisasi spektrum atom dan prinsip ketidakpastian Heisenberg yang semuanya memainkan peran penting dalam ilmu kimia dan fisika Elektron dalam Atom merupakan sumber belajar yang berharga bagi mahasiswa akademisi dan siapa saja yang tertarik untuk memperdalam pemahaman mereka tentang ilmu fisika dan kimia Dengan ilustrasi dan contoh contoh praktis buku ini tidak hanya menjelaskan teori teori kompleks tetapi juga menghubungkannya dengan fenomena fenomena alam yang kita temui dalam kehidupan sehari hari

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exposed to chemistry and its applications beyond the classroom This text is perfect for those instructors who are looking for a more advanced general chemistry textbook **Chemistry** Neil D. Jespersen, Alison Hyslop, 2014-01-13 The Student Solutions Manual to accompany Chemistry The Molecular Nature of Matter 7th Edition Jespersen s Chemistry The Molecular Nature of Matter 7th Edition provides readers with the necessary practice support instruction and assessment that is required for learning and teaching the content of a General Chemistry course This text provides the forum for problem solving and concept mastery of chemical phenomena that leads to proficiency and success The Seventh Edition includes revisions to key content coverage areas and concepts and the addition of more Analyzing Solving Multi Concept problems and examples throughout the text An increased emphasis has also been placed on the intimate relationship that exists between structure at the submicroscopic molecular level and the observable macroscopic properties of matter Jespersen provides readers with a clear concise and easy to understand General Chemistry resource **Cumulated Index to the Books** ,1999

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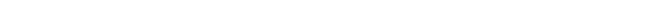
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