

Acids and Bases

Question 1: Choose the correct answer for each of the following questions:

1.1 The element present in all acids...

A oxygen B carbon dioxide
C nitrogen D hydro

1.2 A sample of an acid in a beaker has a pH of 4. An alkali is added to it. What happens to the liquid in the beaker?

A decreases B stays the same
C increases D neutral

1.3 Choose the incorrect combination:

A HNO_3 – Nitric acid. B H_2SO_4 – sulphuric acid
C HCl – Hydrochloric acid. D H_2CO_3 – carbonic acid.

1.4 The colour of blue litmus paper in an acid is ...

A yellow B green
C blue D red

1.5 Which substance is always produced in the reaction between hydrochloric acid and lithium?

A water B hydrogen gas
C oxygen gas D carbon dioxide gas

1.6 Which substance is always produced in the reaction between hydrochloric acid and sodium hydroxide?

A water B hydrogen gas
C oxygen gas D precipitate

1.7 Which formula represents a base?

A KOH B KCl
C CH_3OH D CH_3COOH

1.8 The method of adding an acid to a base to bring about neutralisation is ...

A drop by drop B precipitation
C evaporation D decomposition

[8]

Question 2:

Write the general

reaction for an acid with the following compounds:

2.1 Base:

.....

2.2 Carbonate:

.....

2.3 Metal:

.....

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**Tang Wee Teo,Aik-Ling Tan,Yann
Shiou Ong**

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