

Assessment 1

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Completed on: Tuesday, 23 April 2024, 4:10 PM

Time taken: 30 mins 49 sec(s)

Grade: 68.00 out of 80.00 (77.27%)

Question 1

Correct

Mark: 4.00 out of 4.00

4.00

100% **Correct**

Consider the following name/symbol combinations of elements below. Which one is incorrect?

- a. uranium/U
- b. sulfur/S
- c. nitrogen/N
- d. potassium/K
- e. brom/B Br^2

The correct answer is: brom/B

Question 2

Correct

Mark: 4.00 out of 4.00

4.00

100% **Correct**

How many significant figures does the following number have?

0.0005436

- a. 7
- b. 3
- c. 4 Br^2
- d. 5
- e. 0

The correct answer is: 4

Question 3

Correct

Mark: 4.00 out of 4.00

4.00

100% **Correct**

Calculate the volume (in Litres, L) of a 2.50×10^3 mg metal piece whose density is given as 6.72 g/mL.

- a. $1.68 \times 10^{12} \text{ L}$
- b. $2.69 \times 10^{12} \text{ L}$
- c. $5.95 \times 10^{12} \text{ L}$
- d. $3.72 \times 10^{12} \text{ L}$ Br^2
- e. $1.60 \times 10^{12} \text{ L}$

The correct answer is: $3.72 \times 10^{12} \text{ L}$

Question 4

Correct

Mark: 4.00 out of 4.00

4.00

100% **Correct**

Consider a cube with edge length of 1.2 inches and has a mass of 36 g. Calculate the density in g/cm³ to the correct number of significant numbers?

- a. 2.1 g/cm³
- b. 2.2 g/cm³
- c. 360 g/cm³
- d. 1.3 g/cm³ Br^2
- e. 1.4 g/cm³

The correct answer is: 1.3 g/cm³

Nefe Assessment 3 2 Evaluation Answers

DJ Losen

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