

CHEMISTRY Objective Final Exam Review 10-11

There are 50 questions on the test. The chapters are listed below. Each of the letters represents a concept/idea that you should know or look up.

1. CH 6: Periodic trends, where do they increase & decrease
 - a. Atomic radius & ionic radius
 - b. Electronegativity
 - c. Electron affinity
 - d. Ionization energy
 - e. Reactivity of metals and of non-metals
2. CH 7: Look up properties of elements per group and the names of each group.
 - a. Group 1 very reactive in water; group 7 most reactive non-metals etc.
 - b. Fluorine most reactive in periodic table
 - c. Group 1 will lose e- to become +1
 - d. Elements in same group have same chem properties.
3. CH 8: Ionic compounds and bonds
 - a. Why do atoms bond? How many electrons make them stable?
 - b. How to name ionic compounds, go from name to formula
 - c. What elements make up ionic compounds?
 - d. Properties of ionic compounds
 - e. What is a crystal lattice and the arrangement.
 - f. What is metallic bonding?
4. CH 9: Covalent compounds
 - a. What elements make up covalent compounds?
 - b. Compare ionic vs covalent bonds. An e-neg difference of 1.7 is dividing factor.
 - c. What are pi bonds and sigma bonds?
 - d. Bond length Vs. Bond strength, and Bond dissociation energy.
 - e. What are molecules?
 - f. What are the common diatomic molecules?
 - g. Polar molecules, partial charges, what are they?
5. CH 10: Identify the types of chemical reactions
 - a. Be able to write the products for a double replacement and single replacement reactions.
 - b. Balancing equations
 - c. Writing equations from word names.
6. CH 11: The mole
 - a. Determine the molar mass of a compound
 - b. Calculate the amt of particles, ions, atoms, formula units, molecules in a compound.
 - c. Converting mole \rightarrow mole, mole \rightarrow grams, mole \rightarrow particles and visa versa
 - d. Converting grams \rightarrow mole, grams \rightarrow grams, grams \rightarrow particles visa versa
7. CH 12:
 - a. Stoichiometry conversions using balanced chemical equation
 - b. Mole \rightarrow mole, mole \rightarrow grams, grams \rightarrow grams

Objective Final Review For Chemistry

David Baud

Objective Final Review For Chemistry:

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Boston Journal of Chemistry and Popular Science Review ,1871 The Chemical Review ,1872

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validation in active ... The PDA Technical Report No. 29 – Points to Consider for Cleaning Validation⁴ is also recommended as a valuable guidance document from industry. The following ... Annex 2 Visually clean is an important criterion in cleaning validation. It should be one of the acceptance criteria used on a routine basis. Personnel responsible for ... Points to Consider for Biotechnology Cleaning Validation 49, Points to Consider for Biotechnology Cleaning Validation aligns cleaning validation practices with the life cycle approaches to validation, as enabled by ... What is Cleaning Validation in the Pharmaceutical Industry? Cleaning validation is a process used in the pharmaceutical, biotech, and medical device industries to provide documented evidence that equipment and facilities ... draft working document for comments Sep 21, 2020 — Aspects of cleaning validation and cleaning verification should be considered in quality metrics, with 471 performance indicators identified ... Cleaning Validation Guidelines - A Complete List 2022 [May 2020] Points to consider on the different approaches -including HBEL - to establish carryover limits in cleaning validation for identification of ... Technical Report No. 49 Points to Consider for ... by TF Contributors — Cleaning validation plays an important role in reducing the possibility of product contamination from biopharmaceutical manufacturing equipment. It demonstrates ... Cleaning Validation: Protocol & Guidelines Cleaning validation is a procedure of establishing evidence that cleaning processes for manufacturing equipment prevents product contamination. Cleaning ...