

## Step Step

Start with a square piece of paper. Fold it in half diagonally to form a triangle. Then, fold it in half again to form a smaller triangle. This is the basic shape for the dragon's head.



10

Step 10: A red paper dragon head with a long, curved neck and a small tail.

1

1. Fold the paper in half diagonally to form a triangle. Then, fold it in half again to form a smaller triangle. This is the basic shape for the dragon's head.

1



0

1



## Rupaga's Dragon

1. Fold the paper in half diagonally to form a triangle. Then, fold it in half again to form a smaller triangle. This is the basic shape for the dragon's head.

1



15

15. Fold the paper in half diagonally to form a triangle. Then, fold it in half again to form a smaller triangle. This is the basic shape for the dragon's head.

1



15

15. Fold the paper in half diagonally to form a triangle. Then, fold it in half again to form a smaller triangle. This is the basic shape for the dragon's head.



2

2. Fold the paper in half diagonally to form a triangle. Then, fold it in half again to form a smaller triangle. This is the basic shape for the dragon's head.



2

2. Fold the paper in half diagonally to form a triangle. Then, fold it in half again to form a smaller triangle. This is the basic shape for the dragon's head.



2

2. Fold the paper in half diagonally to form a triangle. Then, fold it in half again to form a smaller triangle. This is the basic shape for the dragon's head.

## Dragon

1. Fold the paper in half diagonally to form a triangle. Then, fold it in half again to form a smaller triangle. This is the basic shape for the dragon's head.



4

1



3

## Summer Dragon

1. Fold the paper in half diagonally to form a triangle. Then, fold it in half again to form a smaller triangle. This is the basic shape for the dragon's head.

2



2



1

## Embossed Dragon

1. Fold the paper in half diagonally to form a triangle. Then, fold it in half again to form a smaller triangle. This is the basic shape for the dragon's head.

7



1



2

## His Dragon's Dragon

1. Fold the paper in half diagonally to form a triangle. Then, fold it in half again to form a smaller triangle. This is the basic shape for the dragon's head.

5

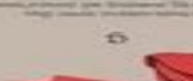


2

## Panther's Dragon

1. Fold the paper in half diagonally to form a triangle. Then, fold it in half again to form a smaller triangle. This is the basic shape for the dragon's head.

6



1

## Flaming Dragon

1. Fold the paper in half diagonally to form a triangle. Then, fold it in half again to form a smaller triangle. This is the basic shape for the dragon's head.



2



7

7. Fold the paper in half diagonally to form a triangle. Then, fold it in half again to form a smaller triangle. This is the basic shape for the dragon's head.



11

11. Fold the paper in half diagonally to form a triangle. Then, fold it in half again to form a smaller triangle. This is the basic shape for the dragon's head.



2

2. Fold the paper in half diagonally to form a triangle. Then, fold it in half again to form a smaller triangle. This is the basic shape for the dragon's head.



1

1. Fold the paper in half diagonally to form a triangle. Then, fold it in half again to form a smaller triangle. This is the basic shape for the dragon's head.

1. Fold the paper in half diagonally to form a triangle. Then, fold it in half again to form a smaller triangle. This is the basic shape for the dragon's head.

# Origami 3d Manual Filetype

**Paul Jackson**



## Origami 3d Manual Filetype:

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web a polar molecule is a water solvent a non polar atom is fat dissolvable it s essential to note here that all together for an answer for structure the atoms must be either both polar or both non polar for instance water is a polar atom as is ethanol when you blend water and ethanol you see a homogeneous arrangement with the two

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11 ci previous question next question

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web the relative electronegativities of the bonded atoms is  $\text{H} < \text{C} < \text{Cl}$  and so the bond moments all point toward the  $\text{Cl}$  end of the molecule and sum to yield a considerable dipole moment the molecules are relatively polar for molecules of high symmetry such as  $\text{BF}_3$  trigonal planar  $\text{CH}_4$  tetrahedral  $\text{PF}_5$  trigonal bipyramidal and  $\text{SF}_6$

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