

# Intro to Science and Macromolecules

There isn't one way to design an experiment! However, this is the **normal** sequence:

1. Ask a question
2. Conduct background research
3. Construct a hypothesis
4. Test your hypothesis in an experiment
5. Analyze data
6. Draw conclusions and communicate them

**Independent Variable:** what the experimenter will deliberately change or manipulate in the investigation

- Usually on the x-axis of a graph

**Dependent Variable:** what changes in response to the independent variable

- On the y-axis of a graph
- what is measured

**Experimental Group:** group that is being experimented on. The group where you change the IV

**Control Group:** group used for comparison with your experimental groups. This is the "normal" group.

**Controlled:** other aspects of an experiment that are held the same

**Macromolecules:** large molecules measured in the DMV is discussed only by the IV

1. Made of cells
2. Have DNA/RNA
3. Reproduce
4. Use energy
5. Maintain Homeostasis
6. Respond to stimuli
7. Grow/ develop
8. Evolve (as a population)

## Macromolecules

Name	Function	Structure	Properties	Examples	Characteristics
Carbohydrates	Supply energy, structure		Water-soluble, energy storage	Glucose, Fructose, Sucrose	DMV 1.4.2.1 1.4.2.2 1.4.2.3 1.4.2.4 1.4.2.5
Proteins	Structural, enzymes, hormones, antibodies		Water-soluble, heat stable, catalytic, structural	Enzymes, Antibodies, Hormones	DMV 1.4.2.6 1.4.2.7 1.4.2.8
Lipids	Storage, insulation, structure		Water-insoluble, heat stable	Fats, Oils	DMV 1.4.2.9 1.4.2.10
Nucleic acids	Information storage, protein synthesis		Acidic, heat stable	DNA, RNA	DMV 1.4.2.11 1.4.2.12 1.4.2.13
Water	Supply, waste removal, temperature regulation		Universal solvent, polar	Water	DMV 1.4.2.14 1.4.2.15 1.4.2.16 1.4.2.17 1.4.2.18

### Basic Chemistry:

Reactants = what you start with

Products = what you make

Protonic: positive charge

Neutrons: no charge

Electrons: negative charge

Polar molecule: unequal

sharing of electrons (one side is positive, other side is negative)

Hydrophilic: loves water

because it is polar

Hydrophobic: hates water

because it is nonpolar

Main elements in living things

CH<sub>2</sub>O (carbon, hydrogen,

nitrogen, oxygen, phosphorus,

water, etc.)

# Organic Macromolecules Study Guide Answers Biology

**James Hall Zimmerman, Sophie E.  
Merritt**

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