

## CHAPTER

## STUDY GUIDE

**The Mole****Section 10.1 Measuring Matter***In your textbook, read about counting particles.***In Column B, rank the quantities from Column A from smallest to largest.****Column A**

8.3 mol  
200  
3  
6,000,000,000  
 $6.02 \times 10^{23}$   
dozen  
four moles  
gross  
pair  
ream

**Column B**

1. \_\_\_\_\_  
2. \_\_\_\_\_  
3. \_\_\_\_\_  
4. \_\_\_\_\_  
5. \_\_\_\_\_  
6. \_\_\_\_\_  
7. \_\_\_\_\_  
8. \_\_\_\_\_  
9. \_\_\_\_\_  
10. \_\_\_\_\_

*In your textbook, read about converting moles to particles and particles to moles.***In the boxes provided, write the conversion factor that correctly completes each problem.**

11.  
atoms

$$1.20 \text{ mol Cu} \times \quad = 7.22 \times 10^{23} \text{ Cu}$$

12.  
 $\times 10^{23} \text{ mol CH}_4$

$$9.25 \times 10^{22} \text{ molecules CH}_4 \times \quad = 1.54$$

13.  
mol Xe

$$1.54 \times 10^{24} \text{ atoms Xe} \times \quad = 2.56 \times 10^2$$

14.  
molecules  $\text{F}_2$

$$3.01 \text{ mol F}_2 \times \quad = 1.81 \times 10^{24}$$

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**O García**



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