



CONTENTS

INTRODUCTION	1-8
--------------------	-----

EXPERIMENTS

11-164

CHEMISTRY

1. pH of Solutions	11-25
(A) To find the pH of the following samples by using pH paper/universal indicator :	13
(i) Dilute hydrochloric acid (HCl)	
(ii) Dilute sodium hydroxide (NaOH) solution	
(iii) Dilute ethanoic acid (CH_3COOH) solution	
(iv) Lemon juice	
(v) Water	
(vi) Dilute hydrogen carbonate solution.	
(B) Studying the properties of acids and bases (HCl and NaOH) on the basis of their reaction with :	17
(a) Litmus solution (blue/red)	
(b) Zinc metal	
(c) Solid sodium carbonate.	
2. Types of Chemical Reactions	26-42
(i) To perform and observe the action of water on quick lime and classify the reaction.	27
(ii) To perform and observe the action of heat on ferrous sulphate crystals and classify the reaction.	31
(iii) To perform and observe the reaction of iron nails kept in copper sulphate solution and classify the reaction.	34
(iv) To perform and observe the reaction between sodium sulphate and barium chloride solutions and classify the reaction.	37

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diagrams to show the nature of image formed 4 To trace the path of the rays of light through a glass prism 5 To trace the path of a ray of light passing through a rectangular glass slab for different angles of incidence 6 To study the dependence of potential difference V across a resistor on the current I passing through it and determine its resistance Also plotting a graph between V and I 7 To determine the equivalent resistance of two resistors when connected in series and parallel Chemistry 8 To find the pH of the following samples by using pH paper universal indicator 9 To study the properties of a base dil NaOH Solution and Acid HCl by their reaction with a Litmus solution Blue Red b Zinc metal c Solid sodium carbonate 10 To perform and observe the following reactions and to classify them into a Combination reaction b Decomposition reaction c Displacement reaction d Double displacement reaction i Action of water on quick lime ii Action of heat on ferrous sulphate crystals iii Iron nails kept in copper sulphate solution iv Reaction between sodium sulphate and barium chloride solutions 11 To observe the action of Zn Fe Cu and Al on the following salt solutions a ZnSO_4 aq b FeSO_4 aq c CuSO_4 aq d Al_2SO_4 3 aq Based on the above result to arrange Zn Fe Cu and Al metals in the decreasing order of reactivity 12 To study the following properties of acetic acid ethanoic acid i Odour ii Solubility in water iii Effect on litmus iv Reaction with sodium hydrogen carbonate 13 To study the comparative cleaning capacity of a sample of soap in soft and hard water Biology 14 To study stomata by preparing a temporary mount of a leaf peel 15 To show experimentally that carbon dioxide CO_2 is given out during aerobic respiration 16 To study A Binary fission in Amoeba and B Budding in yeast with the help of prepared slides 17 To identify the different parts of an embryo of a dicot seed pea gram or red kidney beans Report - Educational Research and Innovations Committee, National Council of Educational Research and Training National Council of Educational Research and Training (India), 1979 **Annual Report** National Council of Educational Research and Training (India), 2007 Complete Science Laboratory Manual CBSE For Class 10 Education Experts, 2019-10 **Indian National Bibliography**, 2015-07 *Core Science Lab Manual with Practical Skills for Class X* V. K. Sally, Chhaya Srivastava, Goyal Brothers Prakashan, 2019-01-17 Goyal Brothers Prakashan *Annual Report* India. Department of Information Technology, 2010

Science Lab Manual Class X Follows the Latest CBSE Syllabus and Other State Board Following the CBSE Curriculum. Edugorilla Prep Experts, 2022-07-05 Theoretical knowledge is not sufficient for the overall development of the student's mind So in order to enhance the mental skills of the Class 10 children CBSE has introduced the concept of Science practicals which develops the overall understanding about the topic This Class 10 Science Lab Manual strictly follows the CBSE based curriculum and other State Board patterns Mr Gopi Chandra Gupta and Mr Shivam Tiwari are the authors of this Science Lab Manual which provides complete knowledge of each term concepts and definitions so that students can refer to this manual for overall work Viva Voce questions given in the Class 10 Science lab manual has experiments that aim to test a student's understanding of the related experiment Some investigatory projects have also been included in this Science Lab Manual This Lab Manual has well labeled diagrams which clearly demonstrate the details of the apparatus that are used the

correct way of handling Laboratory apparatus and perform the experiments methodically The basic purpose of this Lab Manual is to provide the students with appropriate guidance so that they can easily understand carry out and document the experiments in the Science Laboratory The Lab Manual comes with practical lessons followed by lab experiments after each chapter projects and Notebook for jotting down all the essential information The aim of this Science Lab Manual is to help Class 10 students to prepare and understand the complex Science concepts through experimental approach by utilizing pedagogically enhanced content of the Lab Manual and with a compact knowledge base for improving their analytical experimental skills and learning abilities

Practical/Laboratory Manual Chemistry Class XI based on NCERT guidelines by Dr. S. C. Rastogi & Er. Meera Goyal Dr. S. C. Rastogi , Er. Meera Goyal, 2020-06-23 An Excellent Book in Accordance with the latest syllabus for Class 11 Prescribed by CBSE NCERT and Adopted by Various State Education Boards

A Basic Laboratory Techniques

- 1 To cut a glass tube or glass rod
- 2 To bend the glass rod at an angle
- 3 To draw a glass jet from a glass tube
- 4 To bore a cork and fit a glass tube into it

B Characterisation and Purification of Chemical Substances

- 1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique
- 2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method
- 3 To prepare crystals of pure potash alum $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$ from the given impure sample
- 4 To prepare the pure crystals of copper sulphate from the given crude sample
- 5 To prepare pure crystals of benzoic acid from a given impure sample

C Measurement of pH Values

- 1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper
- 2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration
- 3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper
- 4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH
- 5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+

D Chemical Equilibrium

- 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions
- 2 To study the shift in equilibrium between $Co(H_2O)_6^{2+}$ and Cl^- ions by changing the concentrations of either of the ions

E Quantitative Analysis

- 1 To prepare M 10 oxalic acid solution by direct weighing method
- 2 To prepare M 10 solution of sodium carbonate by direct weighing method
- 3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid
- 4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution

F Qualitative Analysis

- 1 Analysis of Anions
- 2 Analysis of Cations

G Detection of Elements in Organic Compounds

- 1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test
- 2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test

INVESTIGATORY PROJECTS

A Checking of Bacterial Contamination in Water

- 1 To check the bacterial contamination in drinking water by testing sulphide ions

B Methods of Water Purification

- 1 To purify water from suspended impurities by

using sedimentation 2 To purify water by boiling 3 To purify water by distillation method 4 To purify water by reverse osmosis technique 5 To purify water by GAC method 6 To purify water by bleach treatment 7 To purify water by oxidising agent 8 To purify water by ozone treatment method C Water Analysis 1 To test the hardness of different water samples D Foaming Capacity of Various Soaps 1 To compare the foaming capacity of different washing soaps 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap E Tea Analysis 1 To study the acidity of different samples of tea leaves tea by using pH paper F Analysis of Fruits and Vegetable Juices 1 To analyse the fruit and vegetable juices for the constituent present in them G Rate of Evaporation 1 To study the rate of evaporation of different liquids H Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Log Antilog Table Lab Manual Science Class 09 Neena Sinha, R.Rangarajan, Rajesh Kumar, These Lab Manuals provide complete information on all the experiments listed in the latest CBSE syllabus The various objectives materials required procedures inferences etc have been given in a step by step manner Carefully framed MCQs and short answers type questions given at the end of the experiments help the students prepare for viva voce **International Books in Print** ,1991

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Table of Contents Ncert Class X Science Lab Manual

1. Understanding the eBook Ncert Class X Science Lab Manual
 - The Rise of Digital Reading Ncert Class X Science Lab Manual
 - Advantages of eBooks Over Traditional Books
2. Identifying Ncert Class X Science Lab Manual
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ncert Class X Science Lab Manual
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ncert Class X Science Lab Manual
 - Personalized Recommendations
 - Ncert Class X Science Lab Manual User Reviews and Ratings
 - Ncert Class X Science Lab Manual and Bestseller Lists

5. Accessing Ncert Class X Science Lab Manual Free and Paid eBooks
 - Ncert Class X Science Lab Manual Public Domain eBooks
 - Ncert Class X Science Lab Manual eBook Subscription Services
 - Ncert Class X Science Lab Manual Budget-Friendly Options
6. Navigating Ncert Class X Science Lab Manual eBook Formats
 - ePub, PDF, MOBI, and More
 - Ncert Class X Science Lab Manual Compatibility with Devices
 - Ncert Class X Science Lab Manual Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ncert Class X Science Lab Manual
 - Highlighting and Note-Taking Ncert Class X Science Lab Manual
 - Interactive Elements Ncert Class X Science Lab Manual
8. Staying Engaged with Ncert Class X Science Lab Manual
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ncert Class X Science Lab Manual
9. Balancing eBooks and Physical Books Ncert Class X Science Lab Manual
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ncert Class X Science Lab Manual
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ncert Class X Science Lab Manual
 - Setting Reading Goals Ncert Class X Science Lab Manual
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ncert Class X Science Lab Manual
 - Fact-Checking eBook Content of Ncert Class X Science Lab Manual
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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