

Multiple Choice: Identify the letter of the choice that best completes the statement or answers the question.

- What scientific (natural) law serves as the basis for balancing chemical equations by requiring that there be no observable change in the quantity of matter during a chemical reaction?
  - Law of Conservation of Energy
  - Law of Multiple Proportions
  - Law of Conservation of Matter
  - Law of Definite Proportions
  - Law of Constant Composition
- Balance the following equation with the smallest whole number coefficients. What is the coefficient for O<sub>2</sub> in the balanced equation?



- 9
- 5
- 15
- 6
- 13

- What is the coefficient for HBr when the following equation is balanced with the smallest whole number coefficients?  
$$\text{Br}_2 + \text{H}_2\text{O} \rightarrow \text{HBr} + \text{HBrO}_3$$
  - 5
  - 7
  - 8
  - 3
  - 6
- Balance the following equation with the smallest whole number coefficients. What is the coefficient for NH<sub>3</sub> in the balanced equation?



- 1
- 3
- 2
- 6
- 4

- Elemental phosphorus is produced from calcium phosphate in the following reaction. What is the coefficient for C when this equation is balanced with the smallest whole number coefficients?  
$$\text{Ca}_3(\text{PO}_4)_2 + \text{SiO}_2 + \text{C} \rightarrow \text{P}_4 + \text{CO} + \text{CaSiO}_3$$
  - 10
  - 3
  - 1
  - 6
  - 4

# McV4U1 Final Exam With Answers

**Doug Irving, Grand Valley Mathematics  
Association**

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