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Steven M. Rice



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Technology and Oligopoly Capitalism Luis Suarez-Villa, 2023-04-14 Technology and Oligopoly Capitalism is a major contribution to our understanding of how technology oligopolies are shaping America's social economic and political reality. Technology oligopolies are the most powerful socioeconomic entities in America. From cradle to grave, the decisions they make affect the most intimate aspects of our lives: how we work, what we eat, our health, how we communicate, what we know and believe, whom we elect, and how we relate to one another and to nature. Their power over markets, trade, regulation, and most every aspect of our governance is more intrusive and farther reaching than ever. They benefit from tax breaks, government guarantees, and bailouts that we must pay for and have no control over. Their accumulation of capital creates immense wealth for a minuscule elite, deepening disparities while politics and governance become ever more subservient to their power. They determine our skills and transform employment through the tools and services they create, as no other organizations can. They produce a vast array of goods and services with labor, marketing, and research that are more intrusively controlled than ever, as workplace rights and job security are curtailed or disappear. Our consumption of their products and their capacity to promote wants is deep and far reaching, while the waste they generate raises concerns about the survival of life on our planet. And their links to geopolitics and the martial domain are stronger than ever, as they influence how warfare is waged and who will be vanquished. Technology and Oligopoly Capitalism's critical multidisciplinary perspective provides a systemic vision of how oligopolistic power shapes these forces and phenomena. An inclusive approach spans the spectrum of technology oligopolies and the ways in which they deploy their power. Numerous previously unpublished ideas expand the repertory of established work on the topics covered, advancing explanatory quality to elucidate how and why technology oligopolies operate as they do, the dysfunctions that accompany their power, and their effects on society and nature. This book has no peers in the literature in its scope, the unprecedented amount and diversity of documentation, the breadth of concepts, and the vast number of examples it provides. Its premises deserve to be taken into account by every student, researcher, policymaker, and author interested in the socioeconomic and political dimensions of technology in America.

Information Needs of Communities Steven Waldman, 2011-09 In 2009, a bipartisan Knight Commission found that while the broadband age is enabling an info and commun renaissance, local communities in particular are being unevenly served with critical info about local issues. Soon after the Knight Commission delivered its findings, the FCC initiated a working group to identify crosscurrent and trend and make recommendations on how the info needs of communities can be met in a broadband world. This report by the FCC Working Group on the Info Needs of Communities addresses the rapidly changing media landscape in a broadband age.

Contents: Media Landscape, The Policy and Regulatory Landscape, Recommendations, Charts and tables. This is a print on demand report.

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical $\times 100$ what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0 6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion no2 which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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lab writeup the purpose of this lab was to observe where substituents are added on benzene ring way of electrophilic aromatic substitution the electrophilic

5 nitration lab docx lab 5 nitration of bromobenzene - Aug 19 2022

chem 2261 lab 5 nitration of bromobenzene introduction an electrophilic aromatic substitution reaction is an important process that allows for intermediate chemical structures to be used as precursors to pharmaceuticals and other industrially produced chemicals by adding different functional groups into an aromatic ring

lab report nitration of bromomobenzene ipl org - Feb 22 2023

lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group