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SCIENCE LAB MANUAL

Internal Assessment



10



Salient Features

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- Basic Concepts of Experiments
- Lab Experiments with Interactive Session & NCERT Lab Manual Questions
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SAGAR**

Ncert Lab Manual Class 10

ANURAG SINGH



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CBSE MCQs Chapterwise For Term I & II, Class 12, Mathematics Career Point Kota,2021-10-12 Strictly as per the new term wise syllabus for Board Examinations to be held in the academic session 2021 22 for class 12 Multiple Choice Questions based on new typologies introduced by the board Stand Alone MCQs MCQs based on Assertion Reason Case based MCQs Include Questions from CBSE official Question Bank released in April 2021 Answer key with Explanations Sample Paper on the latest pattern of Term 1 exam

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point of the given unknown organic compound and its identification simple laboratory technique 2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method 3 To prepare crystals of pure potash alum $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$ from the given impure sample 4 To prepare the pure crystals of copper sulphate from the given crude sample 5 To prepare pure crystals of benzoic acid from a given impure sample C Measurement of pH Values 1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper 2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration 3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper 4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH 5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+ D Chemical Equilibrium 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions 2 To study the shift in equilibrium between Co^{2+} and Cl^- ions by changing the concentrations of either of the ions E Quantitative Analysis 1 To prepare M 10 oxalic acid solution by direct weighing method 2 To prepare M 10 solution of sodium carbonate by direct weighing method 3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid 4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution F Qualitative Analysis 1 Analysis of Anions 2 Analysis of Cations G Detection of Elements in Organic Compounds 1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test 2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test INVESTIGATORY PROJECTS A Checking of Bacterial Contamination in Water 1 To check the bacterial contamination in drinking water by testing sulphide ions B Methods of Water Purification 1 To purify water from suspended impurities by using sedimentation 2 To purify water by boiling 3 To purify water by distillation method 4 To purify water by reverse osmosis technique 5 To purify water by GAC method 6 To purify water by bleach treatment 7 To purify water by oxidising agent 8 To purify water by ozone treatment method C Water Analysis 1 To test the hardness of different water samples D Foaming Capacity of Various Soaps 1 To compare the foaming capacity of different washing soaps 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap E Tea Analysis 1 To study the acidity of different samples of tea leaves tea by using pH paper F Analysis of Fruits and Vegetable Juices 1 To analyse the fruit and vegetable juices for the constituent present in them G Rate of Evaporation 1 To study the rate of evaporation of different liquids H Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Log Antilog Table

Mathematics Lab Manual Class X According to the Latest CBSE Syllabus and Other State Boards Following the CBSE Curriculum Edugorilla Prep Experts, 2022-07-07 This ultimate Class 10 Mathematics Lab Manual covers the comprehensive resources to build strong

foundations of Mathematics for the students who do not want to limit their knowledge to theories but also want to draw the practicals to develop a better understanding about any complex formulas and theorems. The author of this class 10 Mathematics Lab Manual is Mr Mohit Tripathi who has presented the series of integrated information which is strictly based on the latest CBSE Pattern and other State Boards curricula as well. The transdisciplinary approach of this Lab Manual is to foster the inquisitive mindset, critical thinking and Mathematical aptitude amongst the early learners. The Class 10 Mathematics Lab Manual also includes all the projects and activities as per the latest NCERT curriculum. The theories mentioned with each of the activities in the Lab Manual gives a complete understanding of each concept and formulas. This class 10 Mathematics Lab Manual provides the step by step procedure, observations, calculations, results and Viva Voce questions in the idol format that is followed in the Laboratory. The basic purpose of this lab manual not only eases the approach of the Laboratory courses but also to provide the students appropriate guidance required for carrying out the Experiments in math laboratories. All the Experiments in this manual have been given to conform to a systematic format that includes Theory, Material Required, Procedure, Observation, Result, Precautions, Viva Voce and Suggested Activities. The Viva Voce questions in the Manual are followed by the answers which will be useful for the CBSE and state board students. This Class 10 Mathematics Lab Manual comes with practical lessons followed by Lab activity and attached Notebook for jotting down all the essential information. The aim of this Mathematics Lab Manual is to prepare the Class 10 students using pedagogically enhanced content of the lab manual and with a compact knowledge base for improving their analytical, experimental skills and learning abilities.

Indian National Bibliography, 2015-07

Practical/Laboratory Manual

Science Class 10 based on NCERT Guidelines Dr. J. P. Goel, ,Dr. S. C. Rastogi, ,Dr. Sunita Bhagia ,Er. Meera

Goyal, 2023-03-20

Physics

- 1 To determine the focal length of concave mirror
- 2 To find the focal length of convex lens by two pin method
- 3 To find the image distance for varying object distances in case of a convex lens and drawing corresponding ray diagrams to show the nature of image formed
- 4 To trace the path of the rays of light through a glass prism
- 5 To trace the path of a ray of light passing through a rectangular glass slab for different angles of incidence
- 6 To study the dependence of potential difference V across a resistor on the current I passing through it and determine its resistance. Also plotting a graph between V and I
- 7 To determine the equivalent resistance of two resistors when connected in series and parallel

Chemistry

- 8 To find the pH of the following samples by using pH paper: universal indicator
 - a Dilute NaOH solution
 - b Dilute HCl
- 9 To study the properties of a base: dilute NaOH solution and Acid HCl by their reaction with a Litmus solution
 - a Blue
 - b Red
- 10 To perform and observe the following reactions and to classify them into a Combination reaction, b Decomposition reaction, c Displacement reaction, d Double displacement reaction
 - i Action of water on quick lime
 - ii Action of heat on ferrous sulphate crystals
 - iii Iron nails kept in copper sulphate solution
 - iv Reaction between sodium sulphate and barium chloride solutions
- 11 To observe the action of Zn, Fe, Cu and Al on the following salt solutions:
 - a ZnSO_4 aq
 - b FeSO_4 aq
 - c CuSO_4 aq
 - d Al_2SO_4 3 aq

Based on the above result to arrange Zn Fe Cu and Al metals in the decreasing order of reactivity

12 To study the following properties of acetic acid ethanoic acid i Odour ii Solubility in water iii Effect on litmus iv Reaction with sodium hydrogen carbonate

13 To study the comparative cleaning capacity of a sample of soap in soft and hard water

Biology

14 To study stomata by preparing a temporary mount of a leaf peel

15 To show experimentally that carbon dioxide CO_2 is given out during aerobic respiration

16 To study A Binary fission in Amoeba and B Budding in yeast with the help of prepared slides

17 To identify the different parts of an embryo of a dicot seed pea gram or red kidney beans

Practical/Laboratory Manual

Chemistry Class - XI Er. Meera Goyal, 2021-05-29

1 Basic Laboratory Techniques

1 To cut a glass tube or glass rod

2 To bend the glass rod at an angle

3 To draw a glass jet from a glass tube

4 To bore a cork and fit a glass tube into it

Viva Voce

2 Characterisation and Purification of Chemical Substances

1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique

Viva Voce

2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method

Viva Voce

3 To prepare crystals of pure potash alum $\text{K}_2\text{SO}_4 \cdot \text{Al}_2(\text{SO}_4)_3 \cdot 24\text{H}_2\text{O}$ from the given impure sample

4 To prepare the pure crystals of copper sulphate from the given crude sample

5 To prepare pure crystals of benzoic acid from a given impure sample

Viva Voce

3 Measurement of pH Values

1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper

2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration

3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper

4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH

5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+

Viva Voce

4 Chemical Equilibrium

1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions

2 To study the shift in equilibrium between $\text{Co}(\text{H}_2\text{O})_6^{2+}$ and Cl^- ions by changing the concentrations of either of the ions

Viva Voce

5 Quantitative Analysis

1 To prepare M 10 oxalic acid solution by direct weighing method

2 To prepare M 10 solution of sodium carbonate by direct weighing method

3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid

4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution

Viva Voce

6 Qualitative Analysis

Analysis of Anions

Analysis of Cations

Viva Voce

7 Detection of Elements in Organic Compounds

1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test

2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test

Viva Voce

INVESTIGATORY PROJECTS

1 Checking of Bacterial Contamination in Water

1 To check the bacterial contamination in drinking water by testing sulphide ions

Viva Voce

2 Methods of Water Purification

1 To purify water from suspended impurities by using sedimentation

2 To purify water by boiling

3 To purify water by distillation method

4 To purify water by reverse osmosis technique

5 To purify water by GAC method

6 To purify water by bleach treatment

7 To purify water by

oxidising agent 8 To purify water by ozone treatment method Viva Voce 3 Water Analysis 1 To test the hardness of different water samples Viva Voce 4 Foaming Capacity of Various Soaps 1 To compare the foaming capacity of different washing soaps 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap Viva Voce 5 Tea Analysis 1 To study the acidity of different samples of tea leaves tea by using pH paper Viva Voce 6 Analysis of Fruits and Vegetable Juices 1 To analysis the fruit and vegetable juices for the constituent present in them Viva Voce 7 Rate of Evaporation 1 To study the rate of evaporation of different liquids IViva Voce 8 Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Viva Voce

Practical/Laboratory Manual Chemistry Class XII based on NCERT guidelines by Dr. S. C. Rastogi, Er. Meera Goyal Dr. S. C. Rastogi ,Er. Meera Goyal,2020-06-22

A Surface Chemistry 1 To prepare colloidal solution sol of starch 2 To prepare a colloidal solution of egg albumin 3 To prepare colloidal solution of gum 4 To prepare colloidal solution of aluminium hydroxide $Al(OH)_3$ 5 To prepare colloidal solution of ferric hydroxide $Fe(OH)_3$ 6 To prepare colloidal solution of arsenious sulphide As_2S_3 7 To purify a freshly prepared sol by dialysis 8 To compare the effectiveness of different common oils Castor oil cotton seed oil coconut oil kerosene oil mustard oil in forming emulsions Viva Voce B

Chemical Kinetics 1 To study the effect of concentration on the rate of reaction between sodium thiosulphate and hydrochloric acid 2 To study the effect of temperature on the rate of reaction between sodium thiosulphate and hydrochloric acid 3 To study the rate of reaction of iodide ions with hydrogen peroxide at different concentrations of iodide ions 4 To study the rate of reaction between potassium iodate KIO_3 and sodium sulphite Na_2SO_3 using starch solution as indicatorl Viva Voce C

Thermochemistry 1 Determine the enthalpy of dis solution of copper sulphate $CuSO_4 \cdot 5H_2O$ in water at Room temperature 2 To determine the enthalpy of neutralization of the reaction between HCl and $NaOH$ 3 To determine enthalpy change during the interaction between acetone and chloroform Viva Voce D

Electrochemistry 1 To study the variation of cell potential in $Zn|Zn^{2+}||Cu^{2+}|Cu$ with change in concentration of electrolytes $CuSO_4$ or $ZnSO_4$ at room temperature Viva Voce E

Chromatography 1 To separate the coloured components pigment present in the given extract of leaves and flowers by ascending paper chromatography and find their R_f values 2 To separate the coloured components present in the mixture of red and blue inks by ascending paper chromatography and find their R_f values 3 To separate Co^{2+} and Ni^{2+} ions present in the given mixture by using ascending paper chromatography and determine their R_f values Viva Voce F

Preparation of Inorganic Compounds 1 Preparation of double salt of ferrous ammonium sulphate Mohr s salt from ferrous sulphate and ammonium sulphate 2 To prepare a pure sample of potash alum fitkari 3 Preparation of crystals of potassium ferric oxalate or pottasium trioxlato ferrate III Viva Voce G

Preparation of Organic Compounds 1 Preparation of iodoform from ethyl alcohol or acetone 2 Preparation of acetanilide in laboratory 3 Preparation of b Naphthol aniline dye 4 To prepare a pure sample of dibenzalacetone 5 To prepare a pure sample of p nitro acetanilide Viva Voce H

Tests for the Functional Groups Present in

Organic Compounds Viva Voce I Study of Carbohydrates Fats and Proteins 1 To study simple reactions of carbohydrate 2 To study simple reactions of fats 3 To study simple reactions of proteins 4 To investigate presence of carbohydrates fats and proteins in food stuffs Viva Voce J Volumetric Analysis 1 To prepare 250 ml of M 10 solution of oxalic acid 2 To prepare 250 ml of M 10 solution of ferrous ammonium sulphate 3 Prepare M 20 solution of oxalic acid with its help find out the molarity and strength of the given solution of potassium permanganate 4 Prepare M 20 solution of Mohr's salt using this solution determine the molarity and strength of potassium permanganate solution Viva Voce K Qualitative Analysis Viva Voce INVESTIGATORY PROJECTS 1 To study the presence of oxalate ions in guava fruit at different stages of ripening 2 To study the quantity of casein present in different samples of milk 3 Preparation of soyabean milk and its comparison with natural milk with respect to curd formation effect of temperature etc 4 To study the effect of potassium bisulphite as food preservative at various concentrations 5 To study the digestion of starch by salivary amylase and the effect of pH and temperature on it 6 To study and compare the rate of fermentation of the following materials wheat flour gram flour potato juice and carrot juice 7 To extract essential oils present in saunf aniseed ajwain coriander illaichi cardamom 8 To detect the presence of adulteration in fat oil and butter 9 To investigate the presence of NO_2 in brinjal

CBSE Laboratory Manual
Mathematics Class 9 Divya Malik, 2022-12-13 Once William Paul Thurston said Mathematics is not about numbers equations computations or algorithms it is about Understanding It is a constant effort of NCERT that it puts on its textbooks to promote clearer understanding of concepts in every student As important as theoretical study is Practical study is also essential to prove theories into realities The freshly updated edition of LABORATORY MANUAL Mathematics for class IX has been designed as a complete package to understand all the relevant mathematical experiments in a simple lucid and interactive manner Strictly based on CBSE guidelines each activity includes theory to give deep insights into each concept formula term definition etc Viva Voce questions Notes Suggested Activities Applications etc are accumulated to make concepts clearer in accordance with the curriculum Along with the Activities all the projects listed in NCERT Lab Manual have also been properly covered This book serves as a step by step guide for conducting experiments in such a way that students will not need to refer to any other book for explanations of the concepts An all inclusive guidance book for Maths laboratory activities Coverage of each activity in a simple and lucid manner Detailed and Step by Step procedure for each activity Accurate Figures Illustrations and Diagrams in each activity Detailed Demonstration and Calculations with each activity Notes Applications etc with each activity TABLE OF CONTENT Activity 1 34 List of Projects To Develop Heron's formula for area of a triangle story of Chronology of Indian Mathematicians with their contributions Development of formula for the area of a cyclic quadrilateral With rectangle of given perimeter finding the one with a maximum area and with rectangle of given area finding the ones with least perimeter Knowledge and classification of solid figures with respect to surface areas and volumes Generation of Pythagorean triplets Magic squares with cuboid of given surface area finding the one with maximum volume

and with cuboid of given volumes finding one with least surface area Mathematical designs and patterns Application of algebra in day to day life

Practical/Laboratory Manual Biology Class XI based on NCERT guidelines by Dr. Sunita Bhagia & Megha Bansal Dr. Sunita Bhagia ,Megha Bansal,2020-06-23 An Excellent Book in Accordance with the latest syllabus for Class 11 Prescribed by CBSE NCERT and Adopted by Various State Education Boards

Introduction

- 1 Necessary equipments chemicals and other things for practical work
- 2 General Instructions for practical work
- 3 Special Instructions for practical note book Drawing and Recording
- 4 Special Instructions for spotting

EXPERIMENTS

- 1 To study and describe the flowering plant belonging to family one from each of the families
 - a Solanaceae
 - b Fabaceae
 - c Liliaceae
- 2 To prepare temporary slide of transverse section of dicot monocot stem dicot monocot root
- 3 To study osmosis by potato osmometer
- 4 To study of plasmolysis in epidermal peel of Tradescantia or Rhoeo leaf
- 5 To study the distribution of stomata on the upper and lower surface of a leaf
- 6 To compare the rate of transpiration in upper and lower surface of the leaf
- 7 To test the presence of sugars Glucose Sucrose and Starch proteins and fats and to detect their presence in suitable plant and animal materials
- 8 To study the separation of plant pigments by paper chromatography
- 9 To study the rate of respiration in flower buds leaf tissue and germinating seeds
- 10A To test presence of urea in urine
- 10B To test presence of sugar in urine
- 10C To detect presence of albumin in urine
- 10D To test urine for presence of bile salt

SPOTTING

- 1 Study of compound microscope
- 2 To study the plant specimen and identification with reasons
 - Bacteria
 - Oscillatoria
 - Spirogyra
 - Rhizopus
 - Mushroom
 - Yeast
 - Liverwort
 - Moss
 - Fern
 - Pine
- 3 Study of animal specimens
 - 1 Amoeba
 - 2 Hydra
 - 3 Fasciola Hepatica Liver fluke
 - 4 Ascaris Lumbricoides
 - 5 Hirudinaria Granulosa
 - 6 Pheretima Posthuma
 - 7 Palaemon
 - 8 Bombyx Mori
 - 9 Apis Indica Honeybee
 - 10 Pila Globosa Snail
 - 11 Asterias Starfish
 - 12 Scoliodon Dogfish
 - 13 Labeo Rohita Rohu
 - 14 Rana Tigrina Frog
 - 15 Hemidactylus Lizard
 - 16 Columba Livia Pigeon
 - 17 Oryctolagus Cuniculus Rabbit

4A To study the plant tissues Palisade cells Guard cells Parenchyma Collenchyma Sclerenchyma Xylem and Phloem through prepared slide

4B To study the animal tissue squamous epithelium muscles fibres through prepared slide

4C To study mammalian blood smear by temporary permanent slide

- 5 Study of mitosis in root tip of onion
- 6 Study of different modification in root stem and leaves
- 7 To study and identify different types of inflorescence Racemose and Cymose
- 8 To study imbibition in seed raisins
- 9 To demonstrate that anaerobic respiration take place in the absence of air
- 10 To study human skeleton and joints
- 11 To study the external features of cockroach with help of model or chart

Complete Mathematics Laboratory Manual CBSE For Class 9 Education Experts,2019-10

Lab Manual Science Class 09 Neena Sinha, R.Rangarajan, Rajesh Kumar, These Lab Manuals provide complete information on all the experiments listed in the latest CBSE syllabus The various objectives materials required procedures inferences etc have been given in a step by step manner Carefully framed MCQs and short answers type questions given at the end of the experiments help the students prepare for viva voce

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Ncert Lab Manual Class 10 Introduction

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