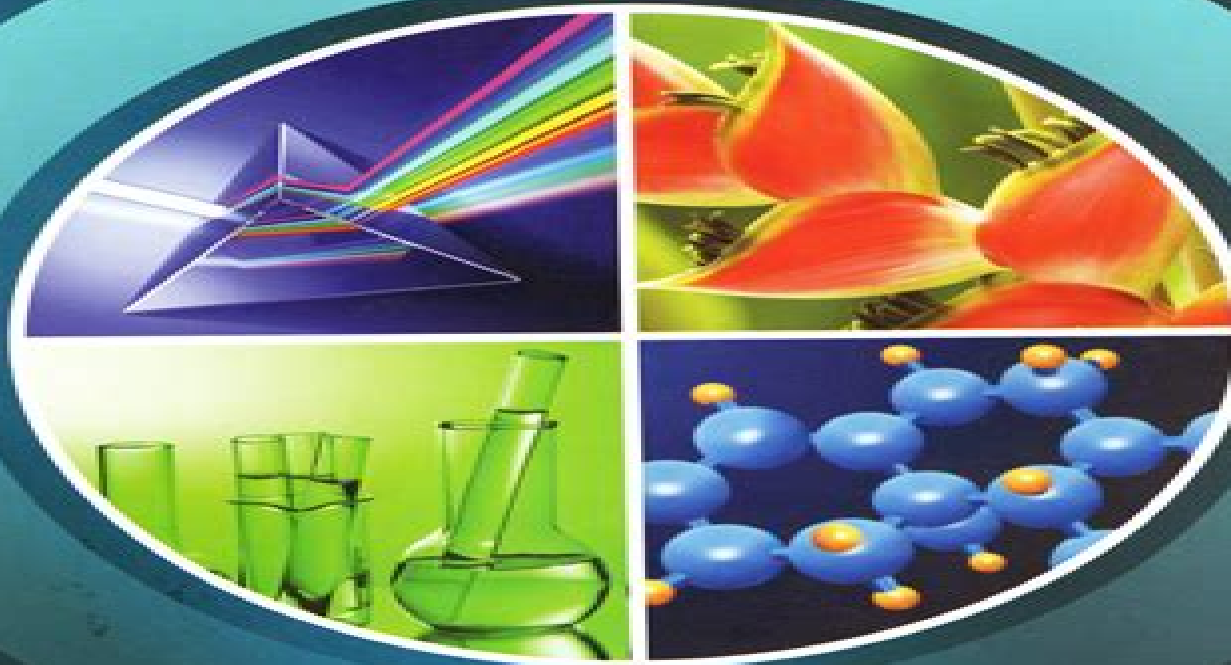




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**R Rangarajan, Neena Sinha, Rajesh
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To find the pH of the following samples by using pH paper universal indicator 9 To studying the properties of a base dil NaOH Solution and Acid HCl by their reaction with a Litmus solution Blue Red b Zinc metal c Solid sodium carbonate 10 To perform and observe the following reactions and to classify them into a Combination reaction b Decomposition reaction c Displacement reaction d Double displacement reaction i Action of water on quick lime ii Action of heat on ferrous sulphate crystals iii Iron nails kept in copper sulphate solution iv Reaction between sodium sulphate and barium chloride solutions 11 To observe the action of Zn Fe Cu and Al on the following salt solutions a ZnSO_4 aq b FeSO_4 aq c CuSO_4 aq d Al_2SO_4 3 aq Based on the above result to arrange Zn Fe Cu and Al metals in the decreasing order of reactivity 12 To study the following properties of acetic acid ethanoic acid i Odour ii Solubility in water iii Effect on litmus iv Reaction with sodium hydrogen carbonate 13 To study the comparative cleaning capacity of a sample of soap in soft and hard water Biology 14 To study stomata by preparing a temporary mount of a leaf peel 15 To show experimentally that carbon dioxide CO_2 is given out during aerobic respiration 16 To study A Binary fission in Amoeba and B Budding in yeast with the help of prepared slides 17 To identify the different parts of an embryo of a dicot seed pea gram or red kidney beans

Practical/Laboratory Manual

Chemistry Class - XI Er. Meera Goyal, 2021-05-29

1 Basic Laboratory Techniques 1 To cut a glass tube or glass rod 2 To bend the glass rod at an angle 3 To draw a glass jet from a glass tube 4 To bore a cork and fit a glass tube into it Viva Voce 2 Characterisation and Purification of Chemical Substances 1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique Viva Voce 2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method Viva Voce 3 To prepare crystals of pure potash alum $\text{K}_2\text{SO}_4 \cdot \text{Al}_2\text{SO}_4 \cdot 24\text{H}_2\text{O}$ from the given impure sample 4 To prepare the pure crystals of copper sulphate from the given crude sample 5 To prepare pure crystals of benzoic acid from a given impure sample Viva Voce 3 Measurement of pH Values 1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper 2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration 3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper 4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH 5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+ Viva Voce 4 Chemical Equilibrium 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions 2 To study the shift in equilibrium between $\text{Co}(\text{H}_2\text{O})_6^{2+}$ and Cl^- ions by changing the concentrations of either of the ions Viva Voce 5 Quantitative Analysis 1 To prepare M 10 oxalic acid solution by direct weighing method 2 To prepare M 10 solution of sodium carbonate by direct weighing method 3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid 4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution Viva Voce 6 Qualitative Analysis Analysis of Anions Analysis of Cations Viva Voce 7 Detection of Elements in Organic

Compounds 1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne s test 2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne s test Viva Voce INVESTIGATORY PROJECTS 1 Checking of Bacterial Contamination in Water 1 To check the bacterial contamination in drinking water by testing sulphide ions Viva Voce 2 Methods of Water Purification 1 To purify water from suspended impurities by using sedimentation 2 To purify water by boiling 3 o purify water by distillation method 4 To purify water by reverse osmosis technique 5 To purify water by GAC method 6 To purify water by bleach treatment 7 To purify water by oxidising agent 8 To purify water by ozone treatment method Viva Voce 3 Water Analysis 1 To test the hardness of different water samples Viva Voce 4 Foaming Capacity of Various Soaps 1 To compare the foaming capacity of different washing soaps 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap Viva Voce 5 Tea Analysis 1 To study the acidity of different samples of tea leaves tea by using pH paper Viva Voce 6 Analysis of Fruits and Vegetable Juices 1 To analysis the fruit and vegetable juices for the constituent present in them Viva Voce 7 Rate of Evaporation 1 To study the rate of evaporation of different liquids IViva Voce 8 Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Viva Voce Concept of Science LAB Manual David King,1915-08-01 Laboratory Manual for Science □ 6

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students The experiments help students verify scientific truths and principles and at the same time expose them to the basic tools and techniques used in scientific investigations Our manuals aim not only to help students better comprehend the scientific concepts taught in their textbooks but also to ignite a scientific quest in their young inquisitive minds **Prep**

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Table of Contents Ncert Core Science Lab Manual Goyal Brothers

1. Understanding the eBook Ncert Core Science Lab Manual Goyal Brothers
 - The Rise of Digital Reading Ncert Core Science Lab Manual Goyal Brothers
 - Advantages of eBooks Over Traditional Books
2. Identifying Ncert Core Science Lab Manual Goyal Brothers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ncert Core Science Lab Manual Goyal Brothers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ncert Core Science Lab Manual Goyal Brothers
 - Personalized Recommendations
 - Ncert Core Science Lab Manual Goyal Brothers User Reviews and Ratings
 - Ncert Core Science Lab Manual Goyal Brothers and Bestseller Lists
5. Accessing Ncert Core Science Lab Manual Goyal Brothers Free and Paid eBooks
 - Ncert Core Science Lab Manual Goyal Brothers Public Domain eBooks
 - Ncert Core Science Lab Manual Goyal Brothers eBook Subscription Services
 - Ncert Core Science Lab Manual Goyal Brothers Budget-Friendly Options
6. Navigating Ncert Core Science Lab Manual Goyal Brothers eBook Formats
 - ePub, PDF, MOBI, and More
 - Ncert Core Science Lab Manual Goyal Brothers Compatibility with Devices
 - Ncert Core Science Lab Manual Goyal Brothers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ncert Core Science Lab Manual Goyal Brothers
 - Highlighting and Note-Taking Ncert Core Science Lab Manual Goyal Brothers
 - Interactive Elements Ncert Core Science Lab Manual Goyal Brothers
8. Staying Engaged with Ncert Core Science Lab Manual Goyal Brothers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Ncert Core Science Lab Manual Goyal Brothers
- 9. Balancing eBooks and Physical Books Ncert Core Science Lab Manual Goyal Brothers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ncert Core Science Lab Manual Goyal Brothers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ncert Core Science Lab Manual Goyal Brothers
 - Setting Reading Goals Ncert Core Science Lab Manual Goyal Brothers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ncert Core Science Lab Manual Goyal Brothers
 - Fact-Checking eBook Content of Ncert Core Science Lab Manual Goyal Brothers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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