

## Acids and Bases

**Question 1:** Choose the correct answer for each of the following questions:

- 1.1 The element present in all acids...
- A oxygen B carbon dioxide  
C nitrogen D hydrogen
- 1.2 A sample of an acid in a beaker has a pH of 4. An alkali is added to it. What happens to the pH of the liquid in the beaker?
- A decreases B stays the same  
C increases D neutral
- 1.3 Choose the incorrect combination:
- A  $\text{HNO}_3$  – Nitric acid. B  $\text{H}_2\text{SO}_4$  – sulphuric acid.  
C  $\text{HCl}$  – Hydrochloric acid. D  $\text{H}_2\text{CO}_3$  – carbonic acid.
- 1.4 The colour of blue litmus paper in an acid is ...
- A yellow B green  
C blue D red
- 1.5 Which substance is always produced in the reaction between hydrochloric acid and lithium?
- A water B hydrogen gas  
C oxygen gas D carbon dioxide gas
- 1.6 Which substance is always produced in the reaction between hydrochloric acid and sodium hydroxide?
- A water B hydrogen gas  
C oxygen gas D precipitate
- 1.7 Which formula represents an acid?
- A  $\text{KOH}$  B  $\text{KCl}$   
C  $\text{CH}_3\text{OH}$  D  $\text{CH}_3\text{COOH}$
- 1.8 The method of adding an acid to a base drop by drop:
- A neutralisation B precipitation  
C evaporation D decomposition

**Quest.** Write the general reaction for an acid with the following compounds:

- 2.1 Base: \_\_\_\_\_
- 2.2 Carbonate: \_\_\_\_\_
- 2.3 Metal: \_\_\_\_\_

# Natural Science Grade 9

**D Keegan**



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