



CHEMISTRY

CHEAT SHEET



NAMING SIMPLE COMPOUNDS

NAMING SIMPLE COMPOUNDS CHEAT SHEET

Naming Compounds with Their Formulas
(Show the ions involved in the compound)



Directions for naming:

1. Write the cation (metal) name first.
2. Write the anion (non-metal) name second, changing the ending to -ide.
3. If the cation is a transition metal, include the oxidation state in Roman numerals in parentheses.
4. If the anion is a polyatomic ion, use its name.
5. If the compound is an acid, use the prefix H- and the name of the anion.
6. If the compound is a base, use the name of the cation and hydroxide.

Binary Compounds

First Element	Anion
H	hydrogen
C	carbon
N	nitrogen
O	oxygen
F	fluorine
Cl	chlorine
S	sulfur
Br	bromine
I	iodine
Al	aluminum
Zn	zinc
Fe	iron
Cu	copper
Pb	lead
Ag	silver

Common Ions and Their Names

Formula	Name
Na ⁺	Sodium
K ⁺	Potassium
Ca ²⁺	Calcium
Mg ²⁺	Magnesium
Al ³⁺	Aluminum
Fe ²⁺	Iron (II)
Fe ³⁺	Iron (III)
Cu ²⁺	Copper (II)
Cu ⁺	Copper (I)
Pb ²⁺	Lead (II)
Pb ⁴⁺	Lead (IV)
Ag ⁺	Silver
Zn ²⁺	Zinc
Sn ²⁺	Tin (II)
Sn ⁴⁺	Tin (IV)

Naming Compounds with a Metal and a Nonmetal



Sodium Chloride

Directions for naming:

1. Write the metal name first.
2. Write the non-metal name second, changing the ending to -ide.

Naming Acids

Formula	Name	Formula	Name
HCl	Hydrochloric acid	HNO ₃	Nitric acid
HBr	Hydrobromic acid	HNO ₂	Nitrous acid
HI	Hydroiodic acid	H ₂ SO ₄	Sulfuric acid
H ₂ SO ₃	Sulfurous acid	H ₂ SO ₄	Sulfuric acid
H ₂ CO ₃	Carbonic acid	H ₂ CO ₃	Carbonic acid
H ₂ SiO ₃	Silicic acid	H ₂ SiO ₃	Silicic acid

Common Names for Specific Compounds

Formula	Common Name	Formula	Common Name
NaCl	Table salt	NaOH	Sodium hydroxide
KCl	Saline	Ca(OH) ₂	Slaked lime
Na ₂ CO ₃	Soda ash	CaCO ₃	Chalk
NaHCO ₃	Baking soda	Na ₂ SO ₄	Glauber's salt

Naming Compounds Study Guide

Susan McMurry



Naming Compounds Study Guide:

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