

Oxidation – Reduction Packet



Oxidation-reduction (redox) reactions are everywhere. They include cellular respiration, combustion reactions and other spontaneous or forced changes involving molecules being broken down or combined. In general, redox reactions transfer electrons from one substance to another.

Oxidation-reduction reactions involve electron transfer and charge balance. When a substance is oxidized, it loses electrons, and the substance it is reduced to gains electrons. Oxidation is the loss of electrons, and reduction is the gain of electrons. The substance that is oxidized is the reducing agent, and the substance that is reduced is the oxidizing agent.

Redox reactions are everywhere. They include cellular respiration, combustion reactions and other spontaneous or forced changes involving molecules being broken down or combined. In general, redox reactions transfer electrons from one substance to another. The substance that is oxidized is the reducing agent, and the substance that is reduced is the oxidizing agent.

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Oxidation And Reduction Packet Answer Key

Mark Steyvers



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