

Inductive Effect:

Permanent effect, across σ bonds
Effect to 3^{rd} carbon, I effect of H-atom
+E effect:



Col NH₂ O Coo To Same CH₃

Pyar Karte He.

-I effect:



Nitrile > Aldehyde > Ketone > Carboxylic

acid > acid chloride > F > Cl > Br > I > CH₃

alcohol > alkene > alkene > alkene

Na Far se Na Rast ke Swide se

Na Heartattack se Na Sote hue

NAK par Acid gira Ac se

AAJ fir C-E ne keya

A APration

Special for Q...

DO Not Play

Disorder number Power of I effect



more acidic less acidic

Application in Stability of intermediate

Acidic & Basic Strength...

Basicity of Aliphatic Amines in Gaseous Phase...



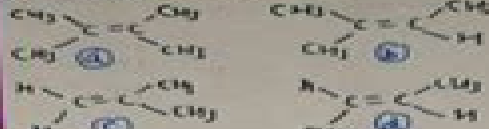
Lower Configuration...

GOC

Application of Hyperconjugation

- Stability of Carbocation
- Stability of Free Radical
- Stability of Alkene
- Stability of alkene

Q stability of alkene:



(A) > (B) > (C) > (D)

Using no. of Hyperconjugation

Stability in C & D is same so:-

(Stability)



1° carbocation



2° carbocation

1° carbocation > 2° carbocation

(Stability)

(JEE 2008 Advance)

Q Hyperconjugation involves overlap of

orbitals:

a) $\sigma - \sigma$

b) $\sigma - \pi$

c) $\pi - \pi$

d) $\pi - \sigma$

Q (JEE 2014)

No. of Hyperconjugative contributing CH



no. of Hyperconjugative contributing CH = 6

Q JEE ADV. 2013

Stability of Carbocation

Resonance

There are certain molecules whose all prop can not be explained by single structure so we draw Resonating Str.

Resonating Str. is called Resonance hybrid. conditions for Resonance:

1. π bond
2. π bond
3. π bond
4. π bond
5. π bond

Rules for more stable Resonating structure:

CH₃NE Mobile CH₃

1. π bond

2. π bond

3. π bond

4. π bond

5. π bond

6. π bond

7. π bond

8. π bond

9. π bond

10. π bond

11. π bond

12. π bond

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28. π bond

29. π bond

30. π bond

31. π bond

32. π bond

33. π bond

34. π bond

35. π bond

36. π bond

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