

## Assessment

## A. Hentriacontane

Molecular Formula —  $C_{31}H_{64}$ Condensed Formula —  $C_3(CH_2)_{29}CH_3$ 

## B. Physical and Chemical Properties of Alkanes

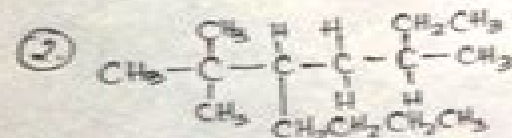
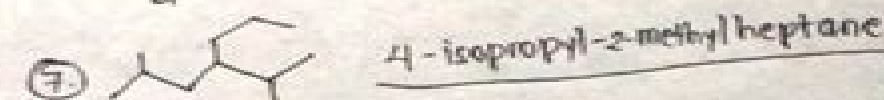
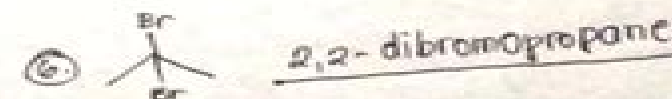
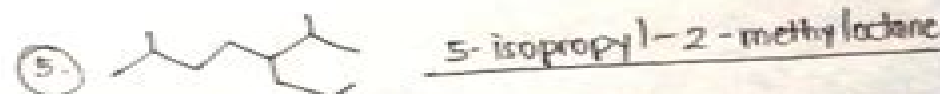
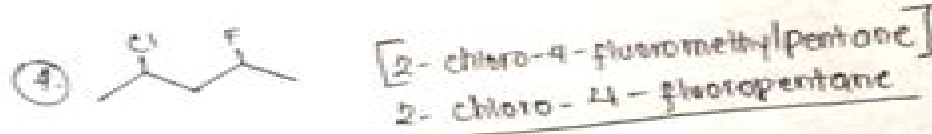
## \* Physical Properties \*

- Low water solubility because of low polarity and hydrophobic characteristic
- Boiling Point increases with the length of the carbon chain.
- Melting Point

## \* Chemical Properties \*

- Paraffins (do not react as most chemicals)
- Combustion (produce  $CO_2$ ,  $H_2O$  and heat)
- Weak van der Waals forces

## C. IUPAC names

2,3-dimethylpentane3-butyl-2,2-dimethyl-5-methylheptane2-methylbutane

# Organic Chemistry Answers

**Michael William Goode**



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