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web the iodination of acetone is convenient because the color of the iodine reactant is readily observable and the change in concentration of 4 fthe iodine can be easily monitored by our spectrophotometer and the reaction proceeds quickly enough for multiple trials but slowly enough to obtain measurements

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web i the iodination of acetone the rate of this reaction is found to depend on the concentration of hydrogen ion in the solution as well as pre sumably on the concentrations of the two reactants

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web video answer the data was given by an point we arrange the dates in ascending order this is the pointing we need to build a stem and plot graph to present quantity data included a graphical input this is the stem part and this is the leaf item yours is

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