

CHAPTER

STUDY GUIDE

The Mole**Section 10.1 Measuring Matter***In your textbook, read about counting particles.***In Column B, rank the quantities from Column A from smallest to largest.****Column A**

8.3 mol
200
3
6,000,000,000
 6.02×10^{23}
dozen
four moles
gross
pair
ream

Column B

1. _____
2. _____
3. _____
4. _____
5. _____
6. _____
7. _____
8. _____
9. _____
10. _____

*In your textbook, read about converting moles to particles and particles to moles.***In the boxes provided, write the conversion factor that correctly completes each problem.**

11.
atoms

$$1.20 \text{ mol Cu} \times \quad = 7.22 \times 10^{23} \text{ Cu}$$

12.
 $\times 10^{23} \text{ mol CH}_4$

$$9.25 \times 10^{22} \text{ molecules CH}_4 \times \quad = 1.54$$

13.
mol Xe

$$1.54 \times 10^{24} \text{ atoms Xe} \times \quad = 2.56 \times 10^2$$

14.
molecules F_2

$$3.01 \text{ mol F}_2 \times \quad = 1.81 \times 10^{24}$$

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**William Hughes Perkins, Karen Jody
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