

## CHEMISTRY FORM THREE NOTES

### CHEMISTRY OF CARBON

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##### A: CARBON

Carbon is an element in Group IV (Group 4) of the Periodic table. It has atomic number 6 and electronic configuration  $2:4$  and thus has four valence electrons (tetravalent). It does not easily ionize but forms strong covalent bonds with other elements including itself.

##### (a) Occurrence

Carbon mainly naturally occurs as:

- (i) allotropes of carbon i.e. graphite, diamond and fullerenes.
- (ii) amorphous carbon in coal, peat, charcoal and coke.
- (iii) carbon(IV) oxide gas accounting 0.03% by volume of normal air in the atmosphere.

##### (b) Allotropes of Carbon

Carbon naturally occur in two main crystalline allotropic forms, carbon-graphite and carbon-diamond

Carbon-diamond	Carbon-graphite
Shiny crystalline solid	Black/dull crystalline solid
Has a very high melting/boiling point because it has a very closely packed giant tetrahedral structure joined by strong covalent bonds	Has a high melting/boiling point because it has a very closely packed giant hexagonal planar structure joined by strong covalent bonds
Has very high density (Hardest known natural substance)	Soft
Abrasive	Slippery
Poor electrical conductor because it has no free delocalized electrons	Good electrical conductor because it has free $4^{\text{th}}$ valency delocalized electrons
Is used in making Jewels, drilling and cutting metals	Used in making Lead-pencils, electrodes in batteries and as a lubricant
Has giant tetrahedral structure	Has giant hexagonal planar structure

##### c) Properties of Carbon

##### (i) Physical properties of carbon

Carbon occur widely and naturally as a black solid

It is insoluble in water but soluble in carbon disulphide and organic solvents.

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web aug 22 2022 the pivotal aim of this personal development plan pdp is to outline leadership skills and competencies vital for a strong leader analyze weaknesses and strengths and offer ways for future leadership development helping to eliminate



the existing drawbacks

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web exclusively available on ivypanda leadership can be understood as the competence to transform visualization into actuality with the help of appropriate planning efficient leaders can be developed and promoted by providing ample prospects for growth and involvement

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web this includes a personal plan of growth as well as a professional plan for improvement this paper will attempt to highlight my personal strengths and weaknesses as leader identify the areas needed for improvement and identify the leadership skills and practices that i will use to become an effective leader