

## Atoms and Isotopes

An atom has a small central \_\_\_\_\_, which contains subatomic particles called \_\_\_\_\_ and \_\_\_\_\_. This is then surrounded by orbiting \_\_\_\_\_.

Complete the following table with the mass and charge of these subatomic particles.

	Relative Mass compared to a proton	Relative Charge
Proton		
Neutron		
Electron		

What is an atom called if it loses or gains electrons?

\_\_\_\_\_

How does an atom become an isotope?

\_\_\_\_\_

\_\_\_\_\_

Label the numbers on the following element, taken from the Periodic Table.

\_\_\_\_\_  
number

Radon  
→ 86

\_\_\_\_\_  
number

Tells us how many \_\_\_\_\_

## P6 – Radioactivity – Triple Physics R

	Change in the Nucleus	Particle emitted and charge
$\alpha$ decay		
$\beta$ decay		
$\gamma$ decay		

### Nuclear Equations

When a nucleus emits an alpha particle the atomic number goes down by \_\_\_\_\_ and the mass number goes down by \_\_\_\_\_.

When a nucleus emits a beta particle the atomic number goes up by \_\_\_\_\_ and the mass number stays the same.

What is the change in the atomic number and mass number when a nucleus emits gamma radiation? \_\_\_\_\_

### Detecting Radiation

A teacher demonstrates a Geiger-Müller tube. It clicks when radiation enters the tube. The teacher looks at his results:

Radioactive source

Which source A, B or C emits Gamma radiation?

### Half-Life

# Ocr Gateway P6 Revision

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