

## Atoms and Isotopes

An atom has a small central \_\_\_\_\_, which contains subatomic particles called \_\_\_\_\_ and \_\_\_\_\_, this is then surrounded by orbiting \_\_\_\_\_.

Complete the following table with the mass and charge of these subatomic particles.

|          | Relative Mass compared to a proton | Relative Charge |
|----------|------------------------------------|-----------------|
| Proton   |                                    |                 |
| Neutron  |                                    |                 |
| Electron |                                    |                 |

What is an atom called if it loses or gains electrons?

How does an atom become an isotope?

Label the numbers on the following element, taken from the Periodic Table.

\_\_\_\_\_ number → Radon 86 / number  
Tells us how many

## P6 – Radioactivity – Triple Physics R

|         | Change in the Nucleus | Particle emitted and charge |
|---------|-----------------------|-----------------------------|
| α decay |                       |                             |
| β decay |                       |                             |
| γ decay |                       |                             |

### Nuclear Equations

When a nucleus emits an alpha particle the atomic number goes down by \_\_\_\_\_ and the mass number goes down by \_\_\_\_\_.

When a nucleus emits a beta particle the atomic number goes up by \_\_\_\_\_ and the mass number stays the same.

What is the change in the atomic number and mass number when a nucleus emits gamma radiation? \_\_\_\_\_

### Detecting Radiation

A teacher demonstrates and a Geiger-Müller tube clicks when radiation passes inside the tube. The teacher says:

Radon → \_\_\_\_\_

Which source A, B, C or D emits Gamma radiation?

Half-Life

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