

Practice Problems for Naming Inorganic Compounds

Write the name (1-25) or formula (26-50) for each of the following inorganic compounds:

1. $\text{Pb}_2(\text{ClO}_3)_4$ _____

2. SnF_2 _____

3. Co_2O_3 _____

4. $\text{As}(\text{ClO}_4)_3$ _____

5. $\text{Na}_2\text{Cr}_2\text{O}_7$ _____

6. $\text{HC}_2\text{H}_3\text{O}_2(\text{aq})$ _____

7. CaF_2 _____

8. NiCr_2O_4 _____

9. $\text{HI}(\text{aq})$ _____

10. SnCl_4 _____

11. P_2O_5 _____

12. Na_2SO_3 _____

13. AsI_3 _____

14. $\text{Zn}(\text{HCO}_3)_2$ _____

15. KMnO_4 _____

16. NBr_3 _____

17. KOH _____

18. Fe_2N_2 _____

19. $\text{Hg}_2(\text{PO}_3)_2$ _____

20. $\text{HNO}_2(\text{aq})$ _____

21. $(\text{NH}_4)_2\text{SO}_4$ _____

22. MgS _____

23. AlPO_4 _____

24. $\text{Ca}(\text{C}_2\text{H}_3\text{O}_2)_2$ _____

25. AgCN _____

26. barium carbonate _____

27. zinc bromide _____

28. nickel(II) chloride _____

29. chromic acid _____

30. disilicon hexafluoride _____

31. lithium fluoride _____

32. carbonic acid _____

33. strontium(II) sulfate _____

34. mercury(I) sulfide _____

35. diiodine pentoxide _____

36. tin(II) acetate _____

37. cobalt(II) chloride _____

38. silver hypochlorite _____

39. sodium phosphate _____

40. ammonium hydrogen sulfate _____

41. iron(II) sulfate _____

42. magnesium nitrite _____

43. copper(II) hydroxide _____

44. hypochlorous acid _____

45. lithium chromate _____

46. tetraphosphorus heptasulfide _____

47. potassium nitrate _____

48. silver perchlorate _____

49. ammonium oxide _____

50. iron(III) chloride _____

Naming Inorganic Compounds Worksheet

M. L. Gillette, Howard A. Neidig



Naming Inorganic Compounds Worksheet:

Rules for Naming Inorganic Compounds International Union of Pure and Applied Chemistry. Committee for the Reform of Inorganic Chemical Nomenclature, Willem Paulinus Jorissen, 1940* Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 *Naming Inorganic Compounds* George I. Sackheim, 1964 **Te HS&T J** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004-02 *Naming Inorganic Compounds* M. L. Gillette, Howard A. Neidig, 1995-02-01 Science Spectrum Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03

Nomenclature of inorganic chemistry, 1957 *How to Name an Inorganic Substance* Sam Stuart, 2013-10-22 How to Name an Inorganic Substance serves as a guide to the use of nomenclature of inorganic chemistry This book contains a few references to the rules for the nomenclature of organic chemistry as well as of inorganic boron compounds This text defines inorganic compounds as substances consisting of combinations of all the elements except those that comprise mainly of certain chains and rings of carbon atoms with defined atoms and groups attached to these skeletal atoms This book presents as well the background principles involved in or related to nomenclature including oxidation number coordination number multiplying affixes enclosing marks and use of italic letters This guide also explains the various types of names used in chemistry including substitutive names simple binary names molecular hydride or ane names and cations derived by proton addition to molecular hydrides This book is a valuable resource for organic and inorganic chemists *Inorganic Nomenclature* Rodney L. Olsen, 1967 **Inorganic Nomenclature** Wesley E. Lingren, 1980 **Properties of Inorganic Compounds** Sidney L. Phillips, 1997-03-27 In recent decades chemists and material scientists have made tremendous advances in the development and application of inorganic compounds especially compounds for commercial applications New compounds have been synthesized new uses for older compounds have emerged and new methods in synthesis and manufacturing have been developed Tracking down critical data about new or newly important inorganic compounds can still be difficult and time consuming Now scientists and researchers have essential information about 5 000 key compounds at their fingertips in Properties of Inorganic Compounds a flexible powerful new electronic database Compounds in the database were carefully selected for their importance in current applications or their potential uses as precursors in preparing materials For each compound the database includes fundamental information including numerical values descriptive data and information on uses Even current industry and research trends are included in this complete resource All this critical data formerly found in many diverse sources is now only a keystroke away System requirements IBM 486 or higher compatible computer with 40MB hard disk 12MB free capacity 4MB RAM VGA Monitor color MS DOS 3.3 or higher Windows 3.1 or higher or Windows 95 external or internal CD ROM drive Will normally run to a lower performance standard on IBM 386 and or less hard disk and RAM capacity than those stated above **Dictionary of Inorganic Compounds, Supplement 4** Jane E. Macintyre, 1996-12-05 A team of editors have selected important and interesting compounds which

have appeared in the literature since publication of volume 8 Volume 9 contains approximately 1200 compounds which are mostly new entries plus a selection of data on compounds already listed in volumes 1-8 This represents literature coverage to mid 1995 and extends the range of the Dictionary of Inorganic Compounds to over 50 000 compounds How to Name an Inorganic Substance, 1977 **How to Name an Inorganic Substance** International Union of Pure and Applied Chemistry. Commission on the Nomenclature of Inorganic Chemistry, 1977 Dictionary of Inorganic Compounds, Supplement 1 Jane E. Macintyre, 1993-07-15 Issued as volume 8 of Dictionary of inorganic compounds **Dictionary of Inorganic Compounds, Supplement 3** Jane E. Macintyre, 1995-10-12 Issued as volume 8 of Dictionary of inorganic compounds *Dictionary of Inorganic Compounds, Supplement 2* Jane E. Macintyre, 1994-11-13 Issued as volume 7 of Dictionary of inorganic compounds *Dictionary of Inorganic Compounds*, 1992 **Dictionary of Inorganic Compounds** Fiona MacDonald, 1992-08-14 The Dictionary of Inorganic Compounds presents fundamental information on more than 42 000 of the most important and useful inorganic compounds each screened for inclusion according to rigorous criteria With its combination of numerical textual and bibliographic data you typically can find all the information you need in this one publication Organized according to empirical name and indexed by name structural type and CAS Registry number each entry includes Compound name synonyms and physical description CAS Registry number Formula and formula weight Structural type with a diagram or description Source or synthesis Stability solubility melting and boiling points sublimations conditions and vapor pressure Hazard toxicity Spectroscopic information References Supplements to the main work available separately provide information on newer compounds and revised data on compounds already listed Indexes in the second and subsequent supplements are cumulative providing quick access to entries in all the supplements from a single index NOMENCLATURE OF INORGANIC CHEMISTRY. INTERNATIONAL UNION OF PURE AND APPLIED CHEMISTRY. COMMISSION ON THE NOMENCLATURE OF INORGANIC CHEMISTRY., 1971

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