

NAMING COVALENT COMPOUNDS PRACTICE WORKSHEET

A. Write the formulas of the following covalent compounds.

- 1) Antimony Tribromide _____
- 2) Hexaboron Silicide _____
- 3) Chlorine Dioxide _____
- 4) Hydrogen Iodide _____
- 5) Iodine Pentafluoride _____
- 6) Dinitrogen Trioxide _____
- 7) Ammonia _____
- 8) Phosphorus Triiodide _____

B. Write the names of the following covalent compounds:

- 1) P_2S_5 _____
- 2) O_2 _____
- 3) SeF_6 _____
- 4) Si_2Br_6 _____
- 5) SCl_4 _____
- 6) CH_4 _____
- 7) B_2Si _____
- 8) NF_3 _____

C. Complete the table with the appropriate information.

Name of the Compound	Formula
Dinitrogen Trisulfide	
	ClO_2
Carbon Diselenide	
Xenon Hexafluoride	
	P_4O_{10}

Naming Covalent Compounds Answers

Bill W. Tillery

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