

# NAMING COVALENT COMPOUNDS PRACTICE WORKSHEET

A. Write the formulas of the following covalent compounds.

- 1) Antimony Tribromide \_\_\_\_\_
- 2) Hexaboron Silicide \_\_\_\_\_
- 3) Chlorine Dioxide \_\_\_\_\_
- 4) Hydrogen Iodide \_\_\_\_\_
- 5) Iodine Pentafluoride \_\_\_\_\_
- 6) Dinitrogen Trioxide \_\_\_\_\_
- 7) Ammonia \_\_\_\_\_
- 8) Phosphorus Triiodide \_\_\_\_\_

B. Write the names of the following covalent compounds:

- 1)  $P_4S_5$  \_\_\_\_\_
- 2)  $O_2$  \_\_\_\_\_
- 3)  $SeF_6$  \_\_\_\_\_
- 4)  $Si_2Br_6$  \_\_\_\_\_
- 5)  $SCl_4$  \_\_\_\_\_
- 6)  $CH_4$  \_\_\_\_\_
- 7)  $B_2Si$  \_\_\_\_\_
- 8)  $NF_3$  \_\_\_\_\_

C. Complete the table with the appropriate information.

Name of the Compound	Formula
Dinitrogen Trisulfide	
	$ClO_2$
Carbon Diselenide	
Xenon Hexafluoride	
	$P_4O_{10}$

# Naming Covalent Compounds Answers

**Bill W. Tillery**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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