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traditional form factors and as parts of cell phones have become ubiquitous over the last decade But for the most part they remain black boxes to the end user and cannot be reprogrammed or modified This has become an obstacle to researchers in the new field of computational photography who want to use the growing computing power of digital cameras to create images no traditional camera could produce This dissertation presents the Frankencamera platform a digital camera system

designed for computational photography The Frankencamera is a fully open fully programmable digital camera which can be easily modified to test out new research ideas The Frankencamera architecture allows for per frame control of the capture process and accurate synchronization of all the components that make up the camera Based on this architecture this dissertation details two hardware platforms the F2 a flexible custom built camera and the Nokia N900 a commercial smartphone Both platforms can be easily programmed at a high level using the FCam API written to embody the Frankencamera architecture Finally this dissertation presents several sample applications for the Frankencamera platform Several of these applications could not have been developed for any existing camera platform and the ease and speed at which they were written show that the Frankencamera platform is a compelling tool for computational photography

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Intelligent Systems and Data Science Nguyen Thai-Nghe, Thanh-Nghi Do, Salem Benferhat, 2025-10-14 This two volume set constitutes the refereed proceedings of the Third International Conference ISDS 2025 held in Can Tho City Vietnam during October 18-19 2025 The 67 full papers and 2 Short papers included in this book were carefully reviewed and selected from 181 submissions They focus on the development of innovative interpretable and explainable data science and artificial intelligence technologies They were organized in topical sections as follows Part I Intelligent Systems and Data Science CCIS 2714 Part II Intelligent Systems and Data Science CCIS 2716

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oct 10 2017 senyawa halogen reaksi halogen dengan logam halogen bereaksi dengan sebagian besar logam menghasilkan senyawa garam na cl_2 nacl 2fe 3cl_2 2fecl_3 sn 2cl_2 sncl_4 mg cl_2 mgcl_2 2al 3cl_2 2alcl_3 2b 3cl_2 2bcl_3 2si 2cl_2 sicl_4 2k cl_2 2kcl

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gas mulia adalah unsur unsur yang terdapat dalam golongan viiia yang memiliki kestabilan yang sangat tinggi dan sebagian ditemukan di alam dalam bentuk monoatomik karena sifat stabilnya unsur unsur yang terdapat dalam gas mulia yaitu helium he neon ne argon ar krypton kr xenon xe radon rn

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oct 16 2023 kerapatan densitas unsur unsur gas mulia juga akan semakin bertambah dari atas ke bawah sifat fisik golongan viiia gas mulia sulit bereaksi bersifat inert karena konfigurasi elektronnya stabil sehingga jarang ditemukan dalam bentuk senyawa gas mulia sedikit larut dalam air kecuali helium dan neon karena ukuran atomnya terlalu kecil

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gas mulia ditemukan di dekat golongan halogen dalam tabel periodik karena unsur gas mulia memiliki konfigurasi elektron yang penuh unsur unsur tersebut tidak reaktif dan senyawanya tidak dikenal akibatnya gas gas ini dikenal dengan gas inert
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gas mulia gas mulia adalah unsur unsur yang terdapat dalam golongan viiia yang memiliki kestabilan yang sangat tinggi dan sebagian ditemukan di alam dalam bentuk monoatomik unsur unsur yang terdapat dalam gas mulia yaitu helium he neon ne argon ar krypton kr xenon xe radon rn

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sifat sifat gas mulia wujud gas mulia unsur gas mulia terdapat sebagai gas tak berwarna yang monoatomik molekul yg terdiri dari 1 atom ini erat kaitannya dengan struktur elektron oktet dan duplet dari gas mulia

kimia unsur gas mulia dan halogen adjiebrotots com - Jul 30 2023

dec 3 2015 gas mulia dan halogen merupakan unsur kimia yang sifatnya sangat bertolak belakang gas mulia dinamakan mulia karena seperti logam mulia sukar sekali bereaksi sementara halogen halos garam artinya adalah pembentuk garam halogen memiliki 6 bilangan oksidasi sehingga mudah sekali berreaksi dengan unsur lainnya gas mulia noble

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pembuatan gas mulia gas helium mempunyai titik didih yang sangat rendah yaitu 268 8 0c sehingga pemisahan gas helium dari gas alam dilakukan dengan cara pengembunan liquefaction sampai gas alam akan mencair sekitar 156 0c dan gas helium terpisah dari gas alam pembuatan ne ar kr dan xe dilakukan dengan proses destilasi bertingkat

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golongan gas mulia dan halogen beberapa golongan unsur memiliki nama khusus seperti golongan viiia gas mulia dan viia halogen golongan tersebut memiliki nama khusus seperti sifat priodik sifat fisik dan sifat kimia kekhas sifat ke 2 golongan tersebut dapat membantu kita unsur unsur yang lain dalam 1 golongan semakin ke atas dan dalam

mengenal macam macam sifat dan unsur gas mulia kimia - Aug 31 2023

oct 7 2021 disebut gas mulia karena semua unsur pada golongan ini berwujud gas dan memiliki konfigurasi elektron yang sangat stabil sehingga akan sangat sulit untuk bereaksi dengan unsur lainnya unsur unsur gas mulia antara lain adalah helium he neon ne argon ar kripton kr xenon xe dan radon rn

sifat fisis dan kimia dari unsur gas mulia dan halogen ppt - May 16 2022

nov 26 2015 1 wujud gas mulia unsur gas mulia terdapat sebagai gas tak berwarna yang monoatomik ini erat kaitannya dengan struktur elektronoktet dan duplet dari gas mulia sedangkan wujud gas pada suhu kamar disebabkan titik

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oct 16 2014 sifat kimia dinamakan gas mulia karena kereaktifannya yang sangat rendah sempat disebut gas inert yang terbantahkan karena sudah ditemukan senyawa gas mulia dan gas langka yang terbantahkan karena argon ternyata mengisi sekitar 0 94 volume atmosfer bumi helium adalah gas mulia yang pertama ditemukan konfigurasi elektron

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apr 16 2017 mempunyai titik didih dan titik leleh yang sangat rendah jari jari atom gas mulia sangat kecil mempunyai elektron valensi 8 khusus untuk helium elektron valensinya 2 sudah stabil sehingga sukar membentuk senyawa dengan unsur

lain dalam tidak pernah di temukan senyawa gas mulia gas mulia terdapat sebagai molekul monoatomik

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fgas mulia dan halogen sejarah gas mulia golongan gas mulia atau golongan viii a adalah unsure unsur yang memiliki delapan elektron valensi dengan konfigurasi elektronik terluar $ns^2 np^6$ unsur unsur tersebut adalah helium he neon

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may 18 2022 dikutip dari buku kimia unsur golongan gas mulia halogen alkali dan alkali tanah yang disusun oleh tantri ambarsari 2020 halogen adalah senyawa yang unsur unsurnya dapat bereaksi dengan logam untuk membentuk garam reaktivitas unsur halogen sangat besar sehingga unsur unsurnya sulit ditemukan di alam